



# The Formation Factor

Jason Weiss  
Oregon State University

April 2nd 2019

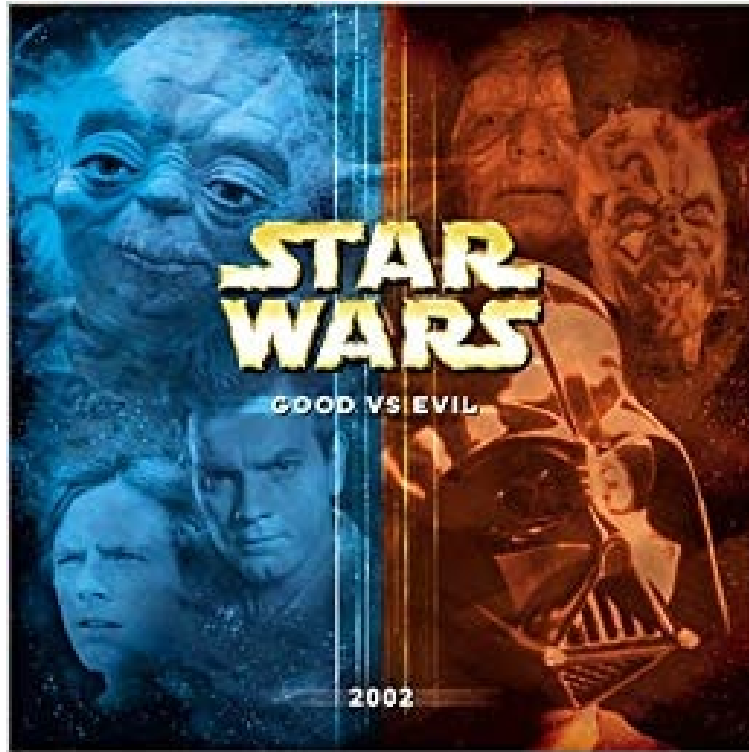
Episode IV

## A NEW HOPE

*It is a period of civil war.  
Rebel spaceships, striking  
from a hidden base, have won  
their first victory against*

STAR  
WAR

# A Challenge at Hand



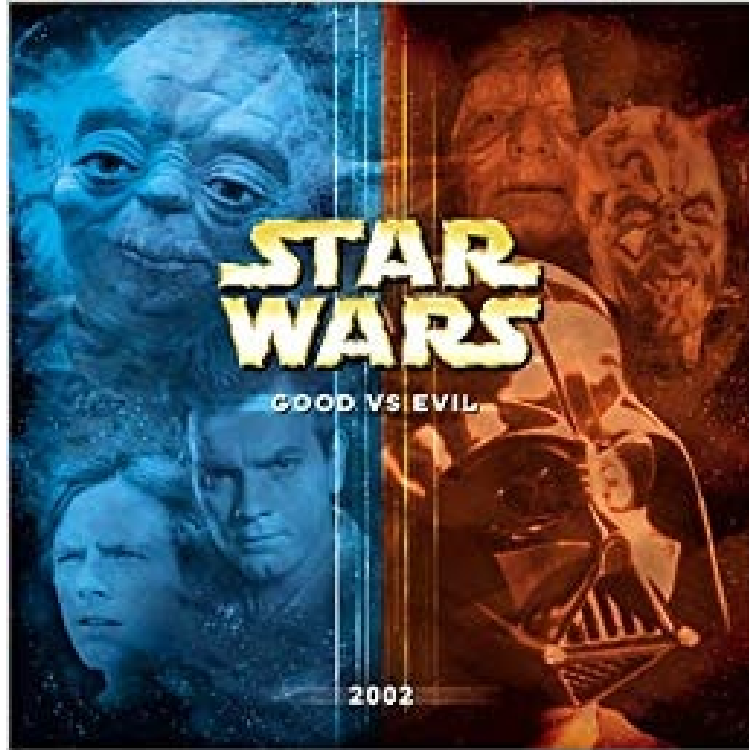
## The Challenge



# A Challenge at Hand



The Hero



The Challenge

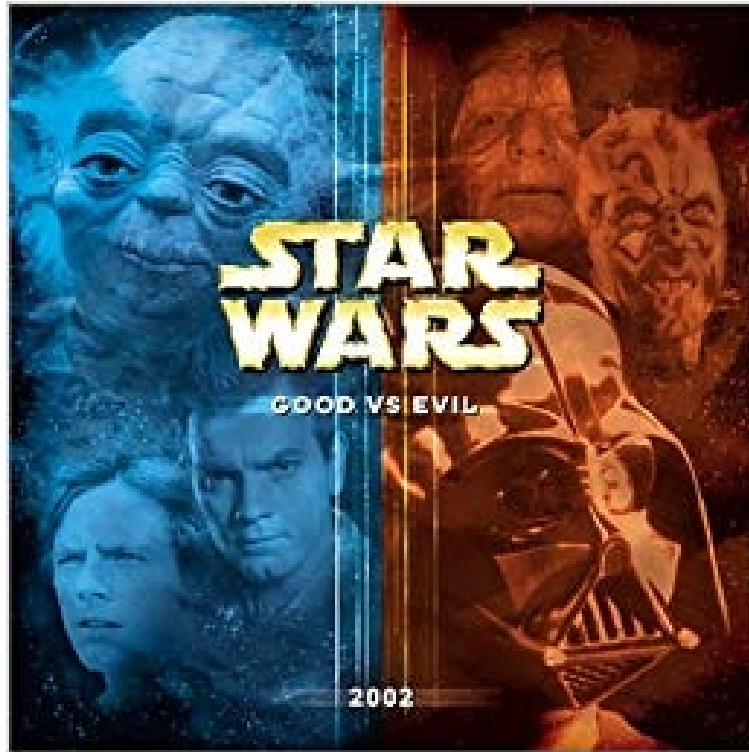
# A Challenge at Hand



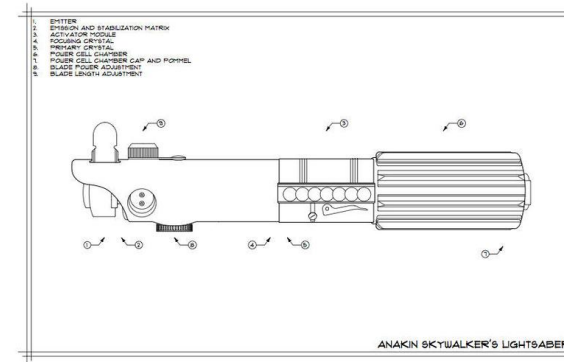
Oregon State University  
College of Engineering



The Hero



The Challenge



The Tools



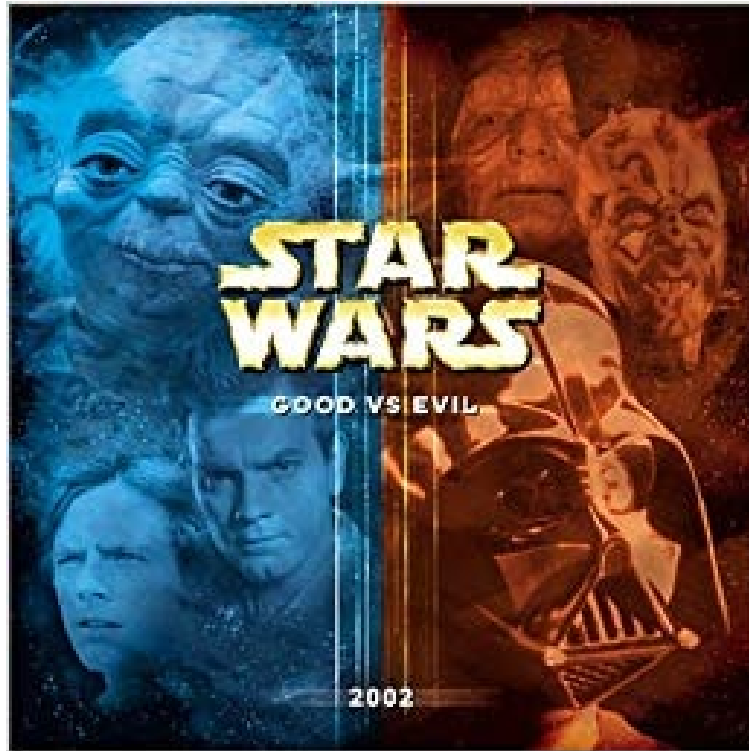
# A Challenge at Hand



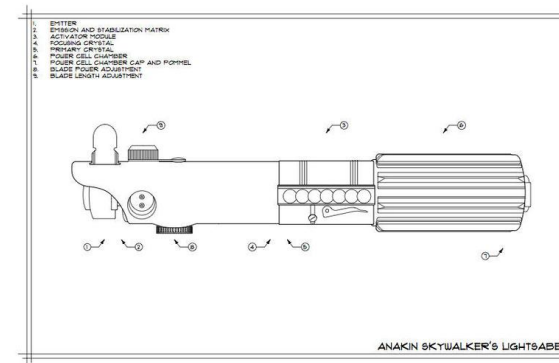
Oregon State University  
College of Engineering



The Hero



The Challenge



Use The Force  
You Must





# Our Challenge at Hand



Oregon State University  
College of Engineering



## Our Challenge

# Our Challenge at Hand



**Our Heros**



**Our Challenge**



# Our Challenge at Hand



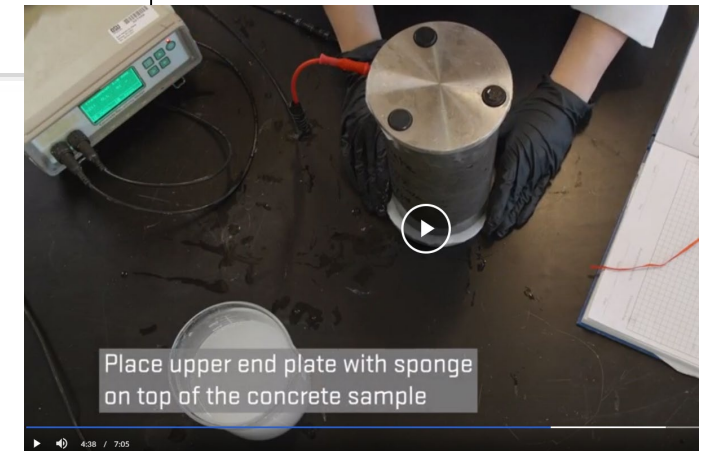
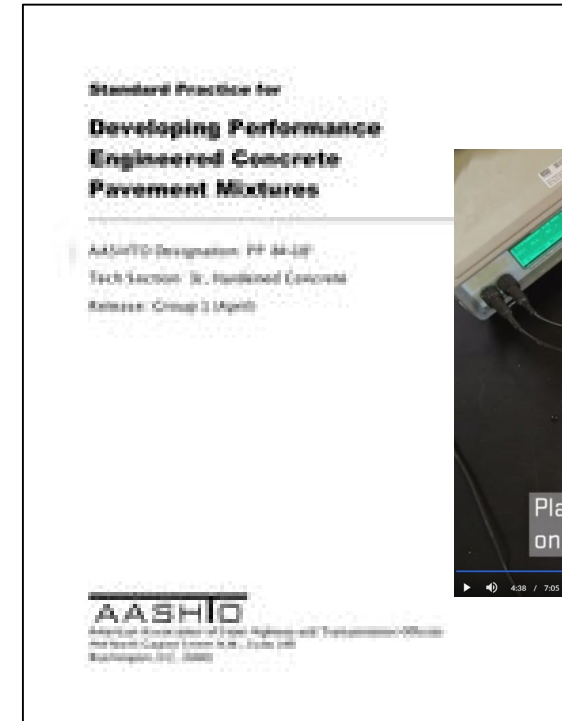
Oregon State University  
College of Engineering



Our Heros



Our Challenge



Our Tools

# Our Challenge at Hand



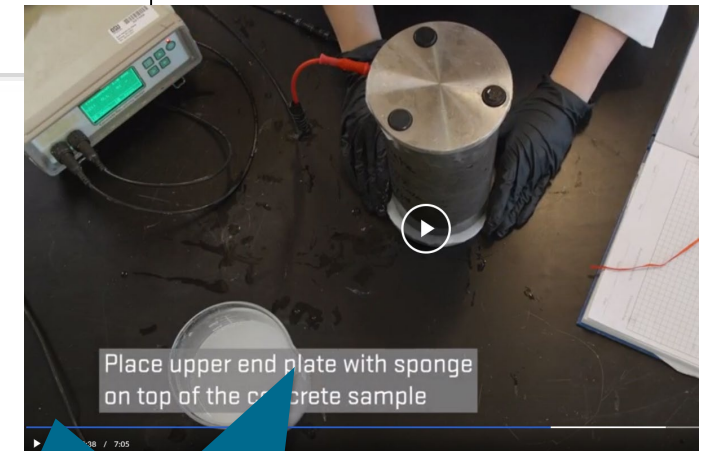
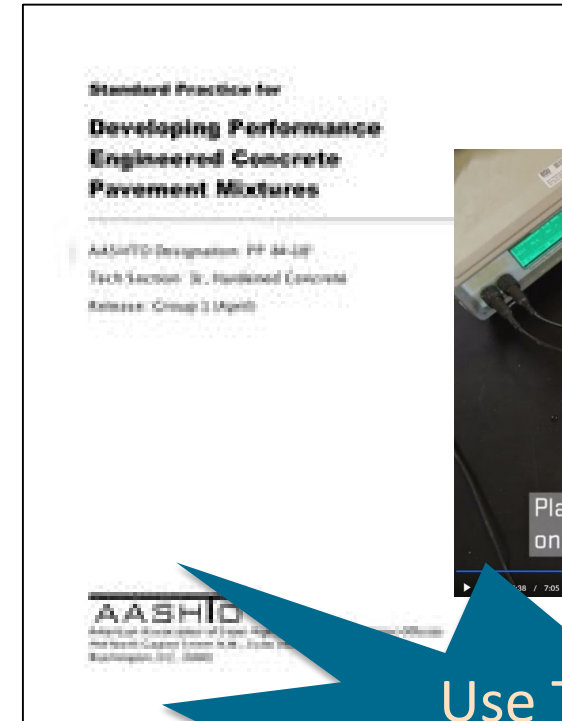
Oregon State University  
College of Engineering



Our Heros



Our Challenge



Use The F-  
Factor You Can

# Before We Can Make Use New Specifications



Oregon State University  
College of Engineering





# Before We Can Make Use New Specifications



Oregon State University  
College of Engineering



# Before implementing new specifications



- Learning how to obtain the Formation Factor
- Lets start with resistivity
- The test is 'easy' but details are important
- Sample conditioning is key
- Identified 4 methods and discussed in Idaho (listed in proposed TP119 revisions)



# I - Geometry is key



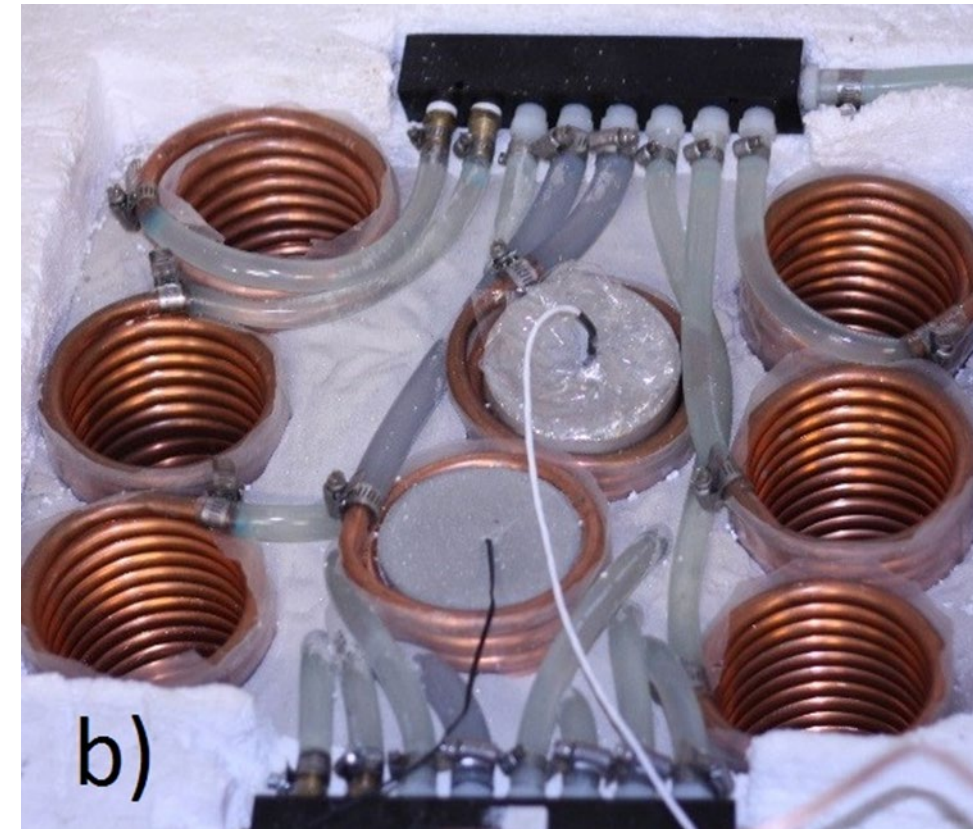
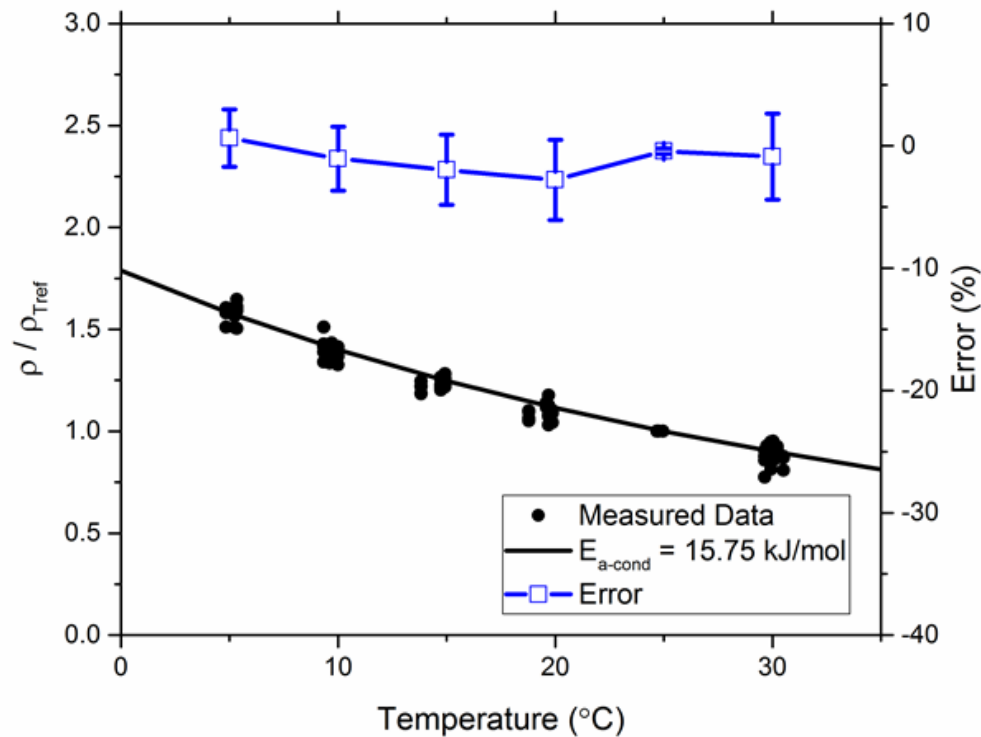
- There are 'three geometries' frequently used
- With proper geometry conditioning, all are similar



- There are two AASHTO test standards; the primary issue in repeatability is related to conditioning
- Temperature, Moisture, Leaching, Degree of Saturation



# II – Temperature is Important



$$\rho_{Tref} = \rho_T \exp \left[ \frac{-E_{a-cond}}{R} \left( \frac{1}{T + 273} - \frac{1}{T_{ref} + 273} \right) \right]$$

# III – Curing and Conditioning



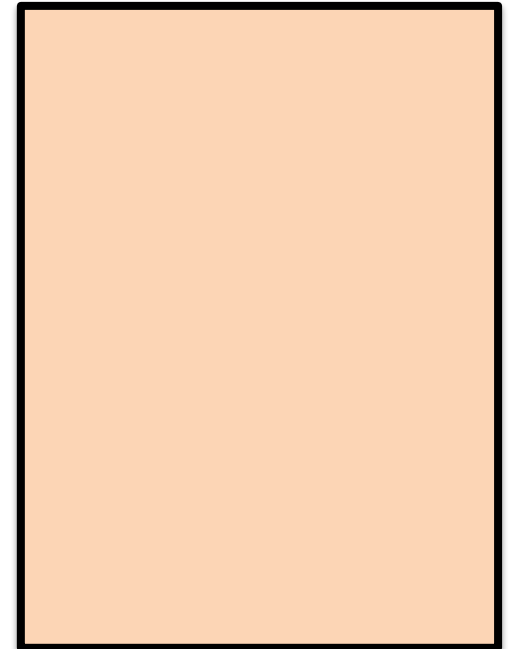
- Option 1 - “The Bucket Test”
  - ~~• Option 2 - Sealed Samples (Not Today)~~
  - ~~• Option 3 – Vacuum Saturation~~
  - ~~• Option 4 – Moist Curing Room~~
- 
- We will start by describing how to perform the tests using simplified procedures

# Option A “Bucket Test”



Oregon State University  
College of Engineering

- Begin with a 5 gallon bucket



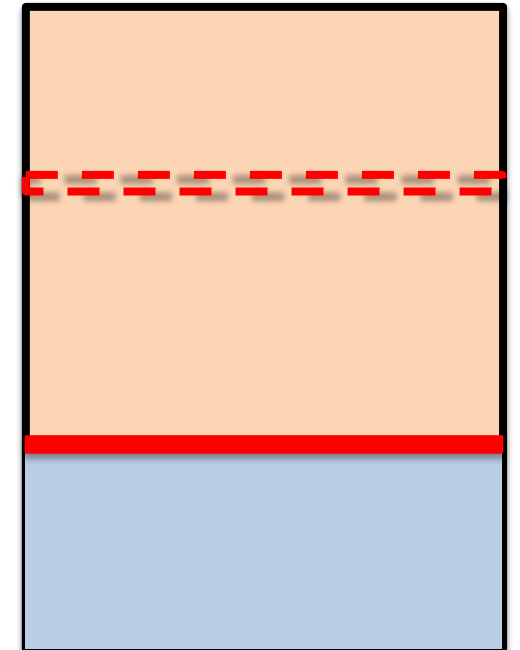


# Option A “Bucket Test”



Oregon State University  
College of Engineering

- Begin with a 5 gallon bucket
- Place a specific volume of fluid into the bucket (the solution to sample ratio is important, place a line on the bucket)

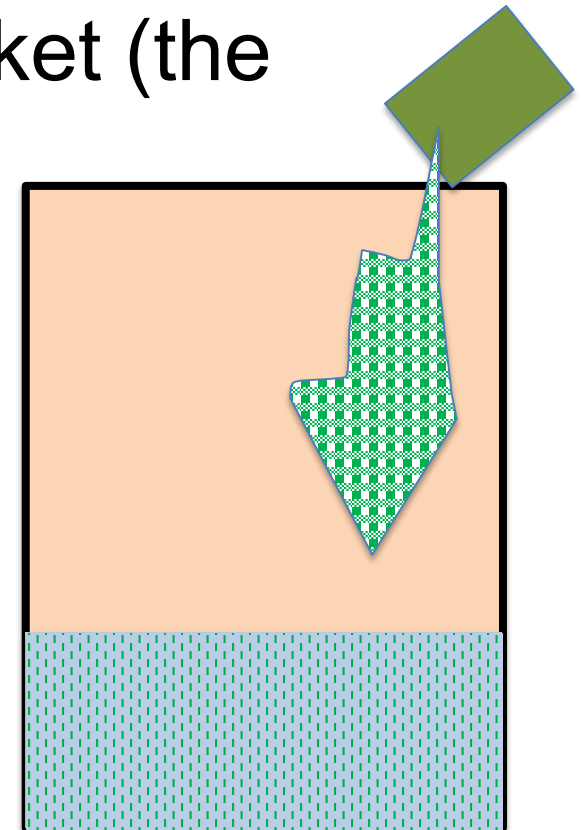


# Option A “Bucket Test”



Oregon State University  
College of Engineering

- Begin with a 5 gallon bucket
- Place a specific volume of fluid into the bucket (the solution to sample ratio is important, place a line on the bucket)
- Place a specified “CH-salt” into the solution (some adjust to the mixture, we suggest selecting a standard value)

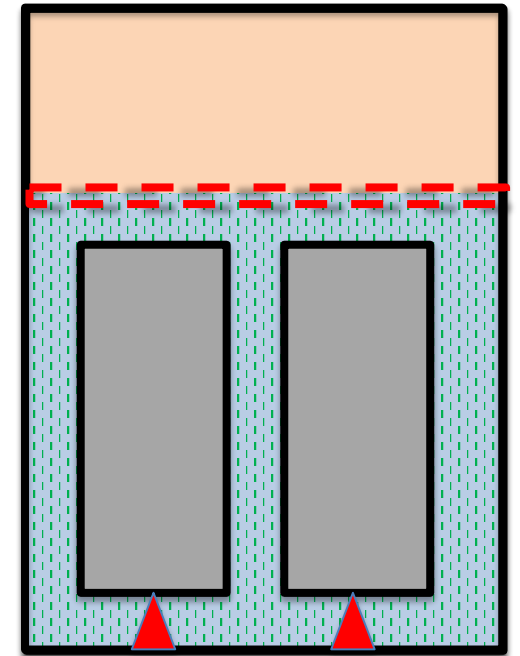


# Option A “Bucket Test”



Oregon State University  
College of Engineering

- Begin with a 5 gallon bucket
- Place a specific volume of fluid into the bucket (the solution to sample ratio is important, place a line on the bucket)
- Place a specified “CH-salt” into the solution (some adjust to the mixture, we suggest selecting a standard value)
- Place samples into the solution to allow the solution to reach the entire sample

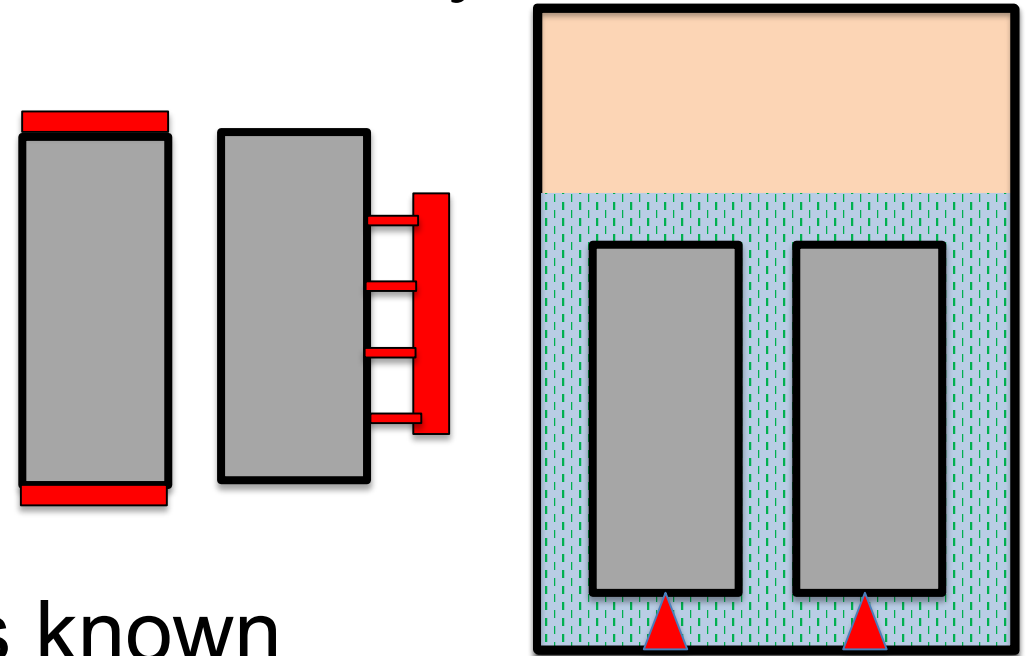




# Option A “Bucket Test”



- At select ages remove sample from the bucket, towel/wash off the surface and perform either the surface resistivity test or the uniaxial bulk resistivity test
- After 5 days in solution the sample is assumed to be in matrix saturation
- This can provide a measure of  $\rho_{\text{measured}}$  or  $F_{\text{matrix}}$
- Conditioning solution  $\rho_{\text{solution}}$  is known



# This is outlined in the AASHTO TP 119 revision



Oregon State University  
College of Engineering



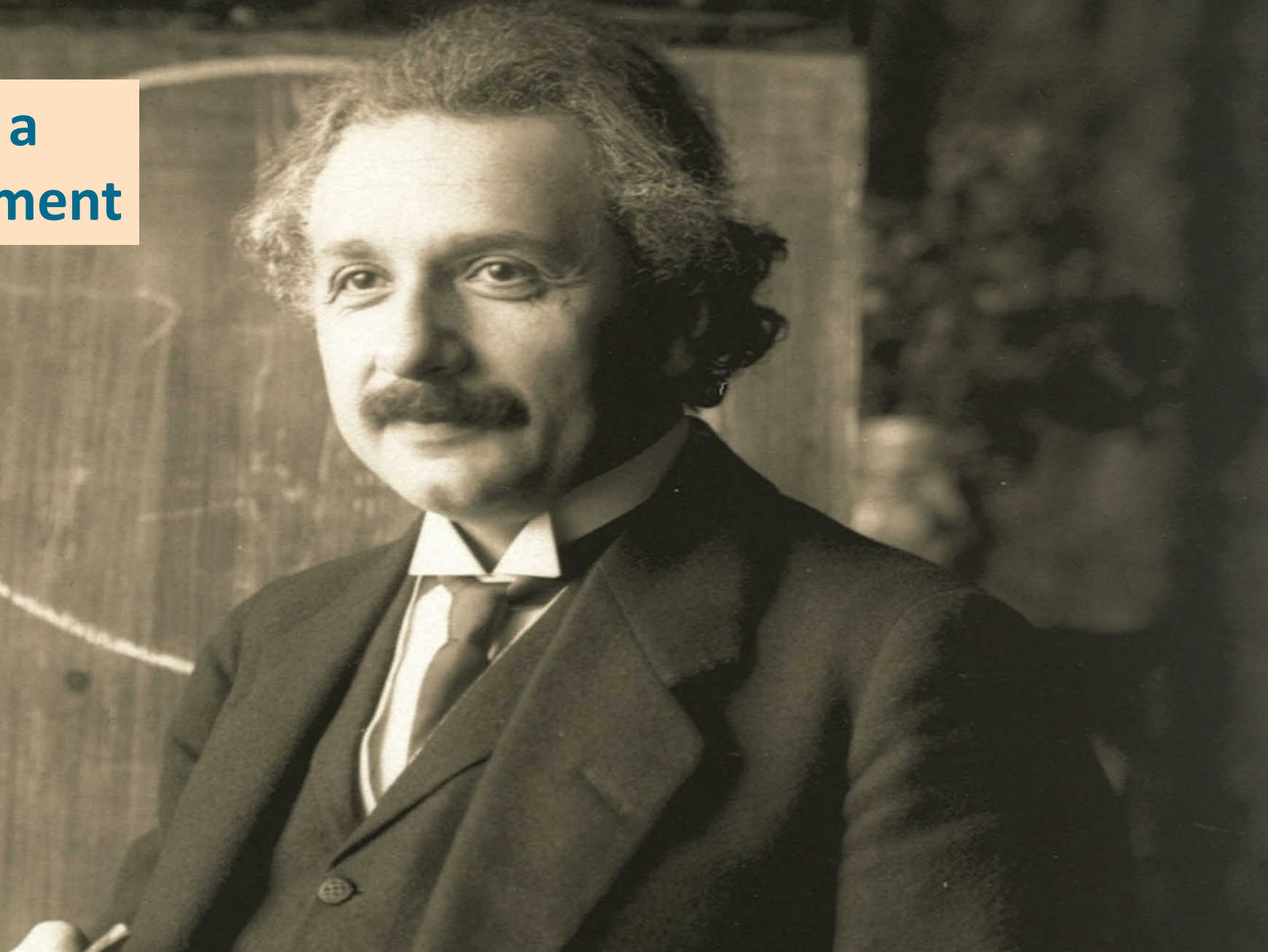
# Things to keep in mind



Oregon State University  
College of Engineering

- The bucket with one solution is a compromise that has been made to make this easy to implement as many in the field are struggling to make pore solutions
- The approaches used several years ago matched the solution in the bucket to the concrete
- This is the best way but has been difficult to implement
- The bucket solution is designed as a 40:1 (approx.) solution exchange
- This is an approximation and is documented in reports

# Lets Perform a Thought Experiment





# Imagine a Chunk of Concrete



Oregon State University  
College of Engineering

Lets Compute  
the RCPT of  
this concrete

Lets have  
an ion to conduct



Gray is a non-  
conductive solid

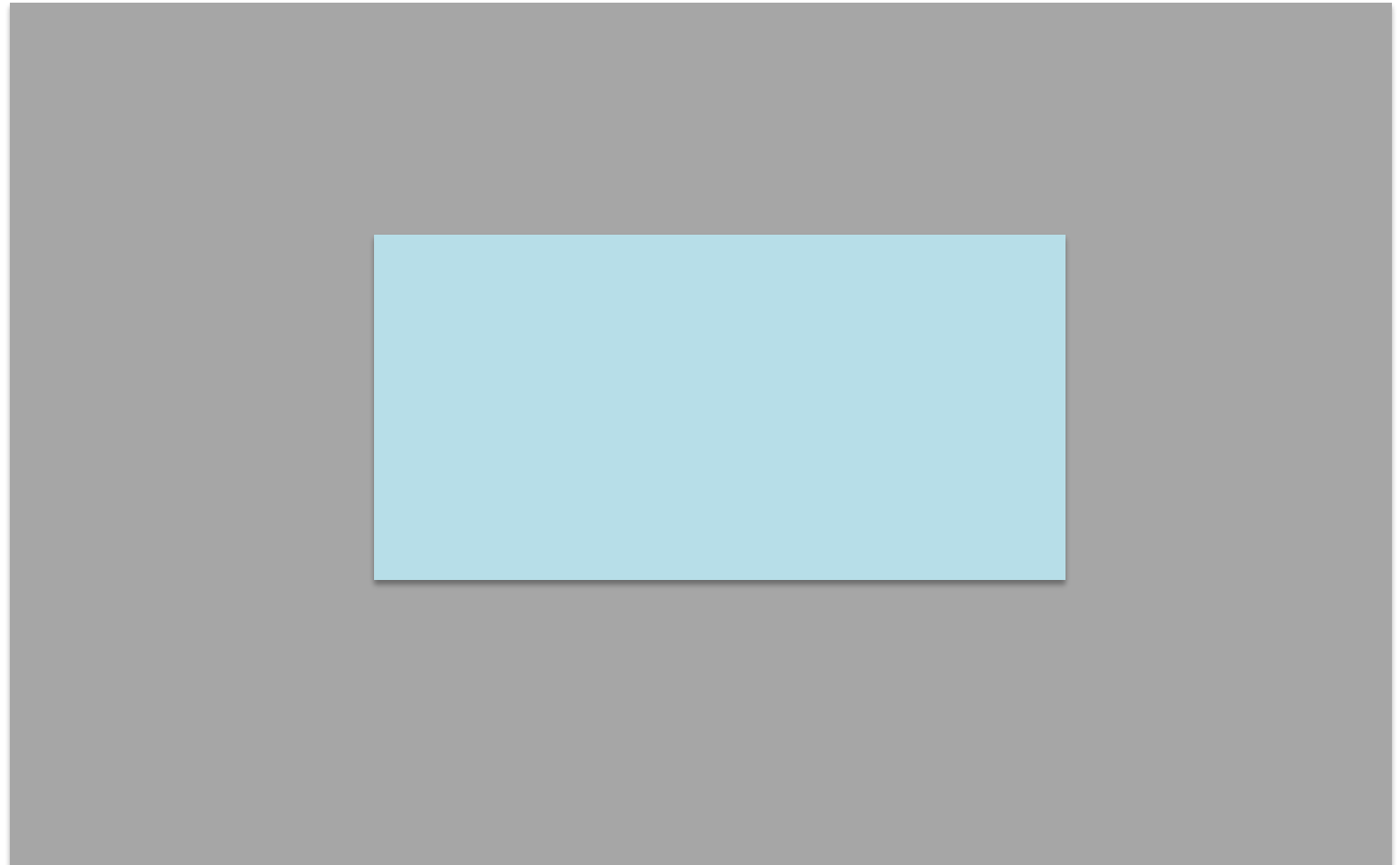


# Imagine a Chunk of Concrete



Oregon State University  
College of Engineering

- 20% Porosity ( $\phi$ )



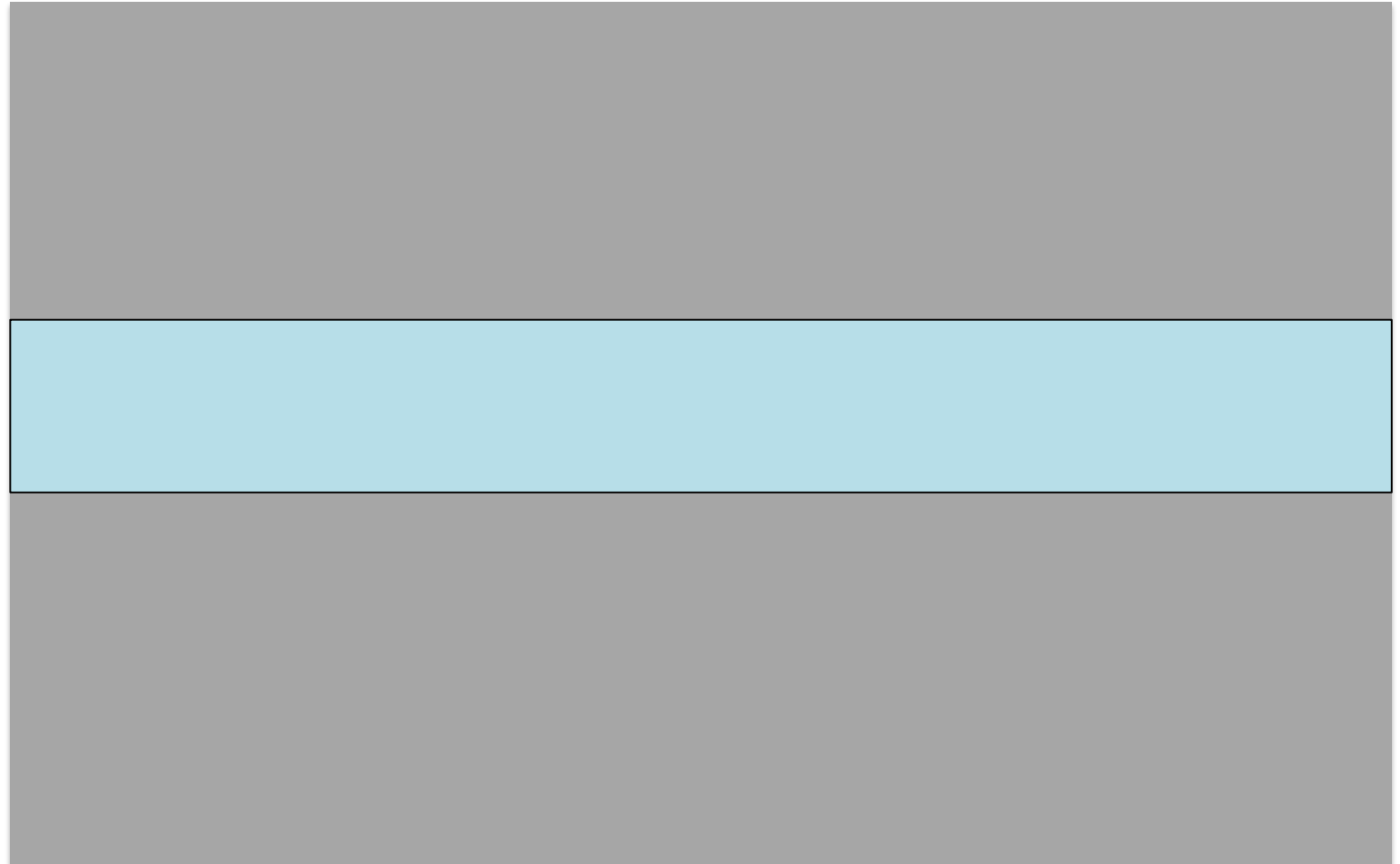
- RCPT = 0

# Imagine a Chunk of Concrete



Oregon State University  
College of Engineering

- 20% Porosity ( $\phi$ )
- 1 Connectivity ( $\beta$ )

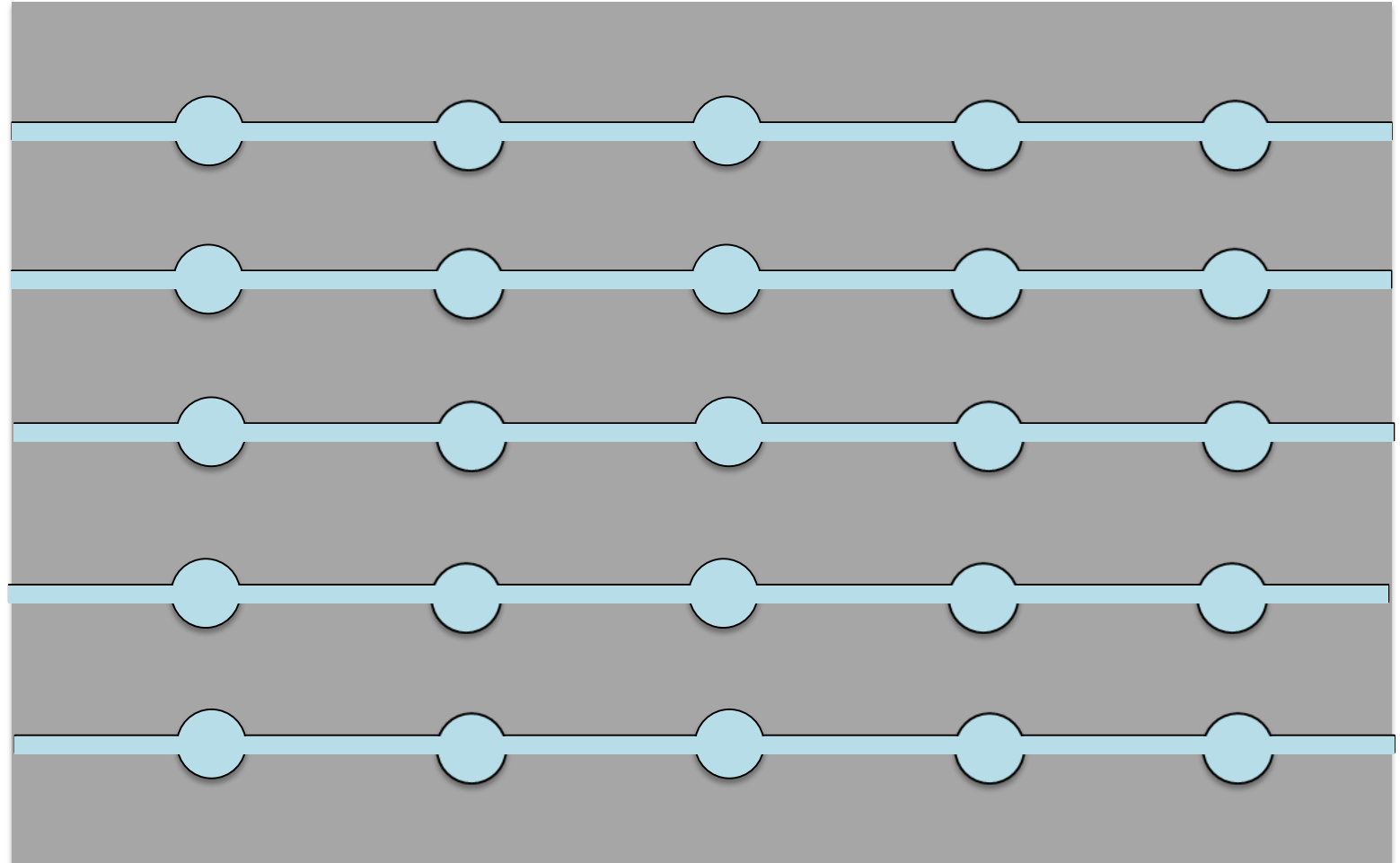


- RCPT = 376,054

# Imagine a Chunk of Concrete



- 20% Porosity ( $\phi$ )
- 0.1 Connectivity ( $\beta$ )



- RCPT = 3,760

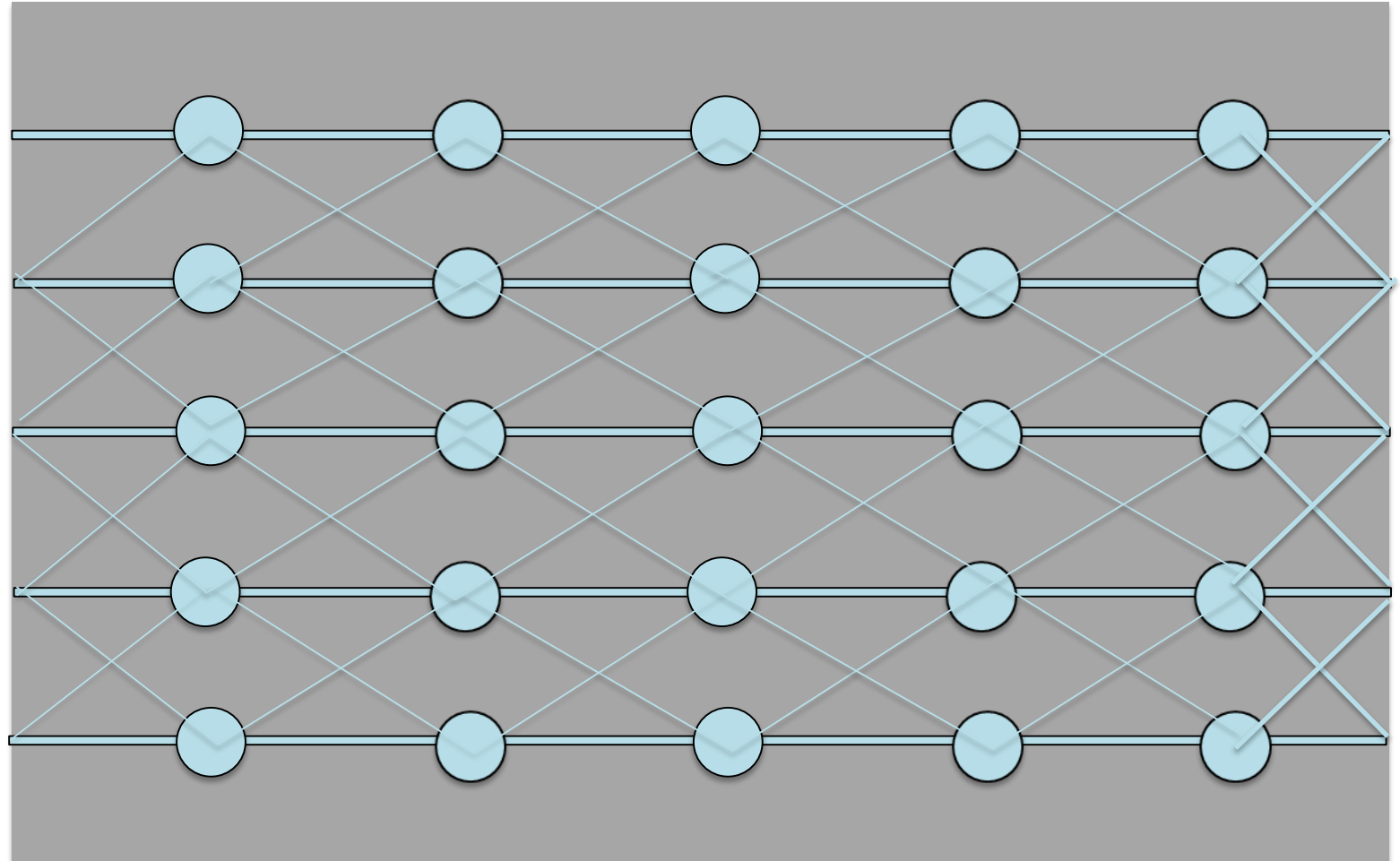


# Imagine a Chunk of Concrete



- 20% Porosity ( $\phi$ )
- 0.004 for a value of Connectivity ( $\beta$ )

Cl<sup>-</sup>

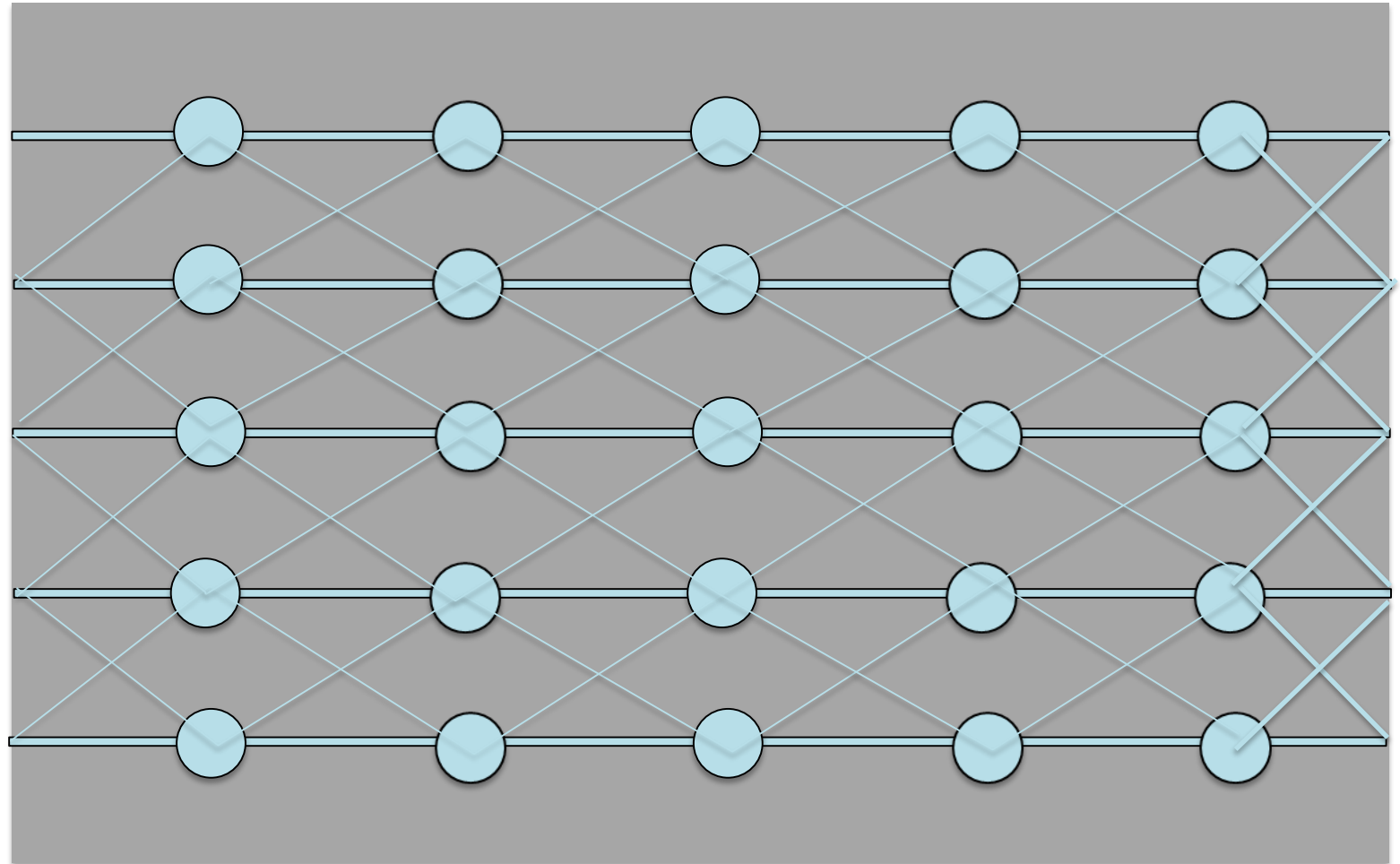


- RCPT = 1500

# Change Alkali Content



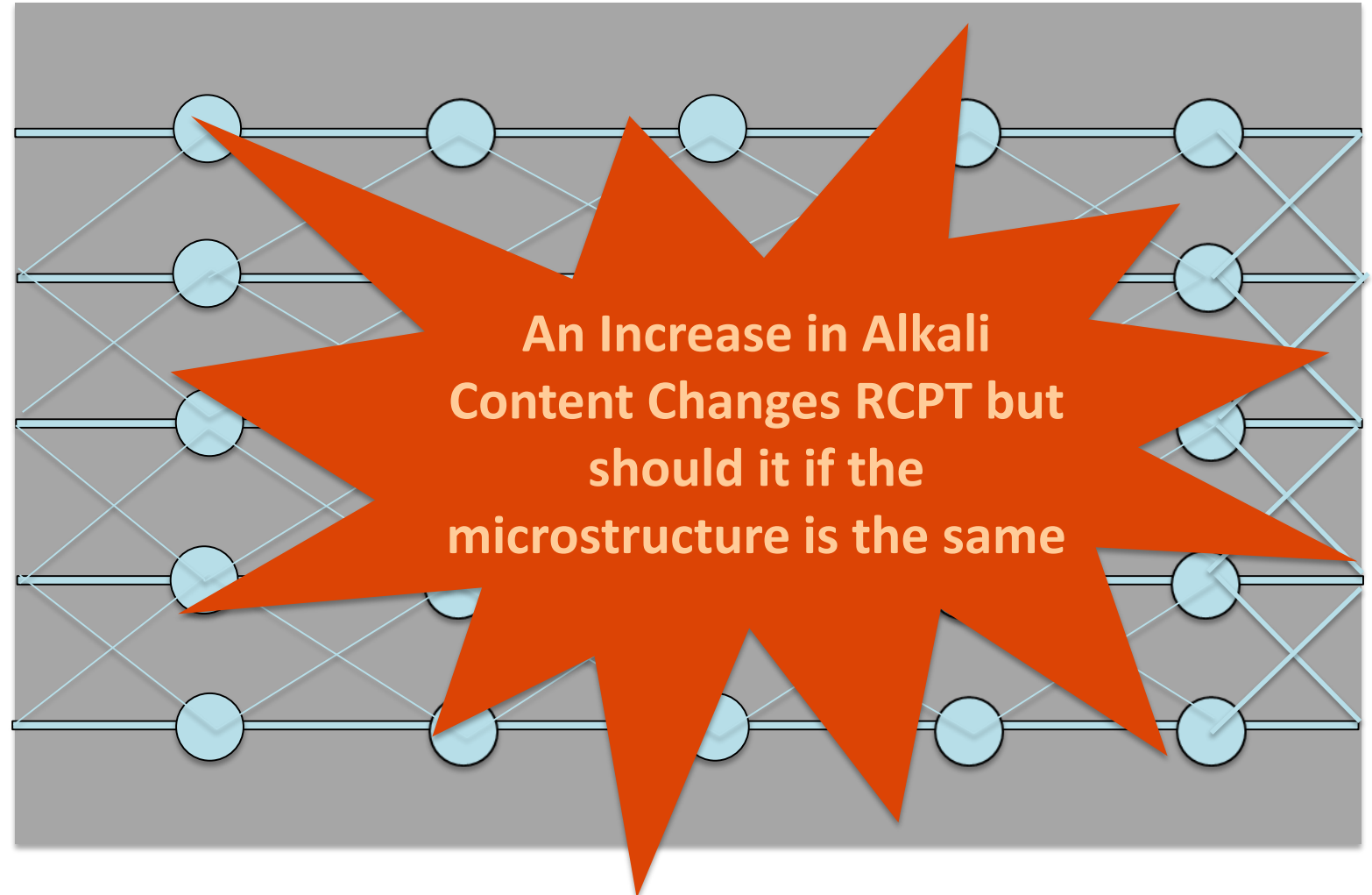
- 20% Porosity ( $\phi$ )
- 0.004 for a value of Connectivity ( $\beta$ )
- Double the alkali content of the cement
- RCPT = 3008



# Change Alkali Content



- 20% Porosity ( $\phi$ )
- 0.004 for a value of Connectivity ( $\beta$ )
- Double the alkali content of the cement
- RCPT = 3008



# Change Alkali Content



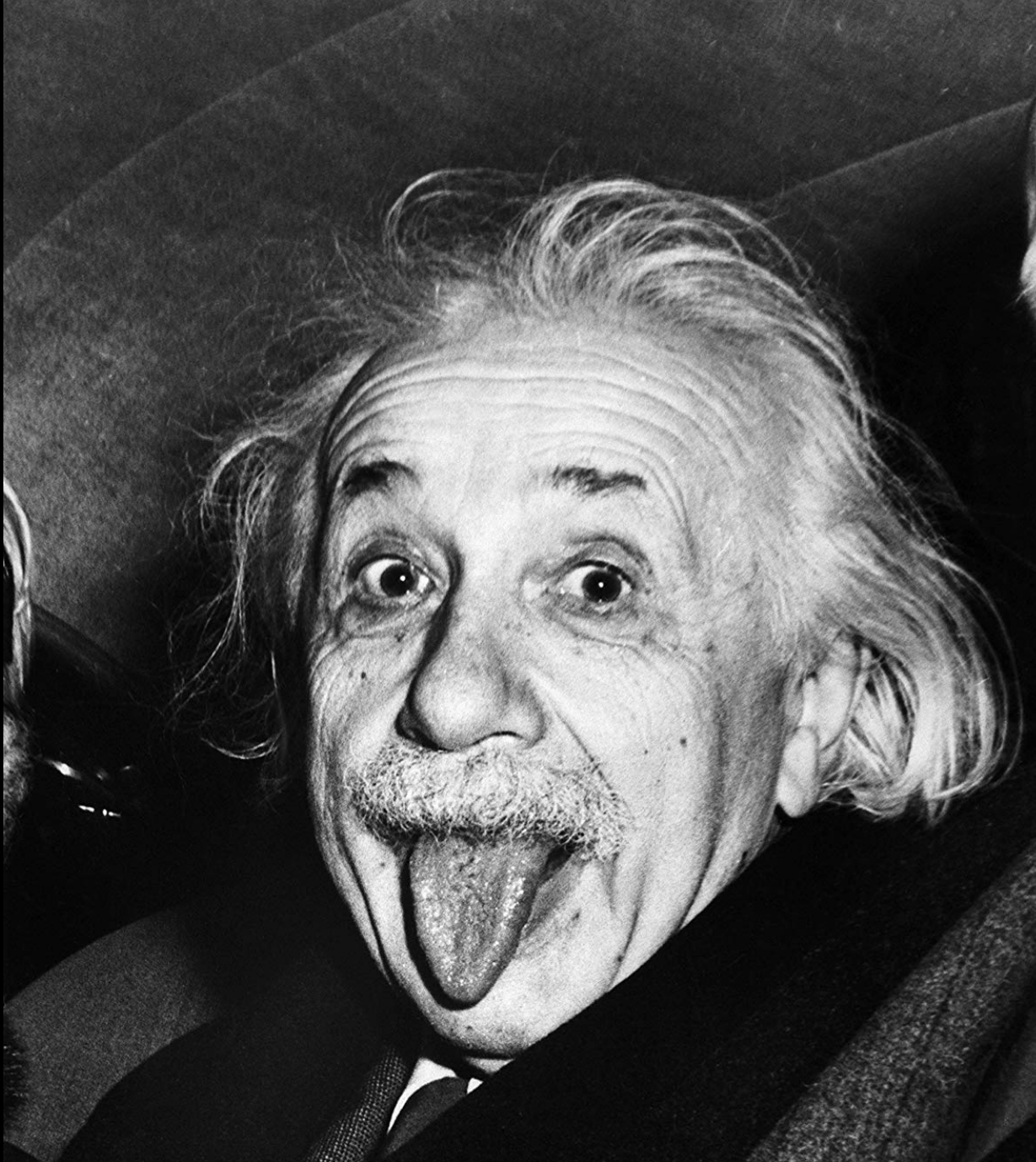
- 20% Porosity ( $\phi$ )
- 0.004 for a value of



An Increase in Alkali  
Content Changes RCPT but  
should it if the  
microstructure is the same

- RCPT = 3008





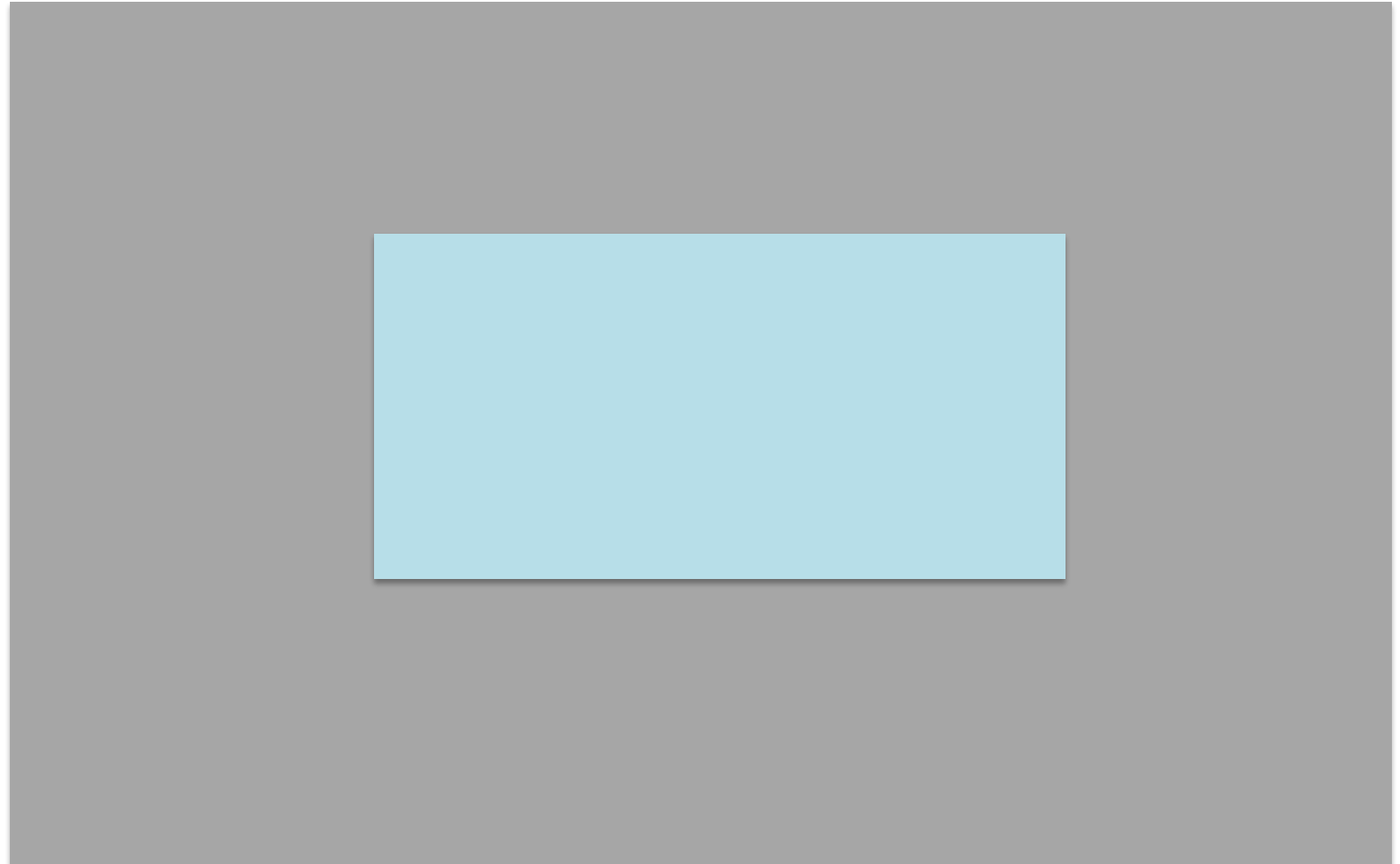
**Lets Perform a  
Thought Experiment  
But Lets Use  
the Formation  
Factor This Time**

# Imagine a Chunk of Concrete



Oregon State University  
College of Engineering

- 20% Porosity ( $\phi$ )



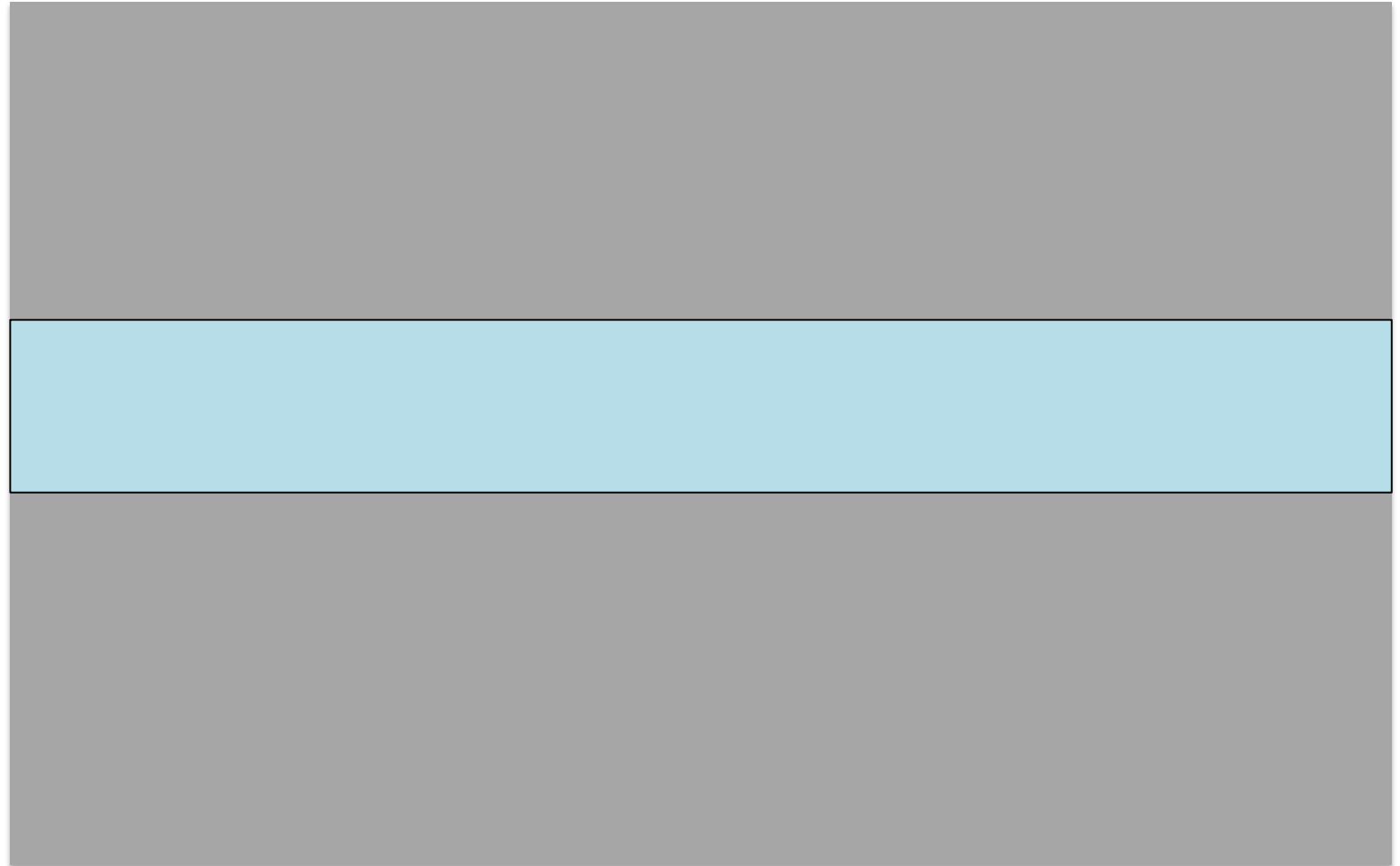
- $F = \infty$

# Imagine a Chunk of Concrete



Oregon State University  
College of Engineering

- 20% Porosity ( $\phi$ )
- 1 Connectivity ( $\beta$ )

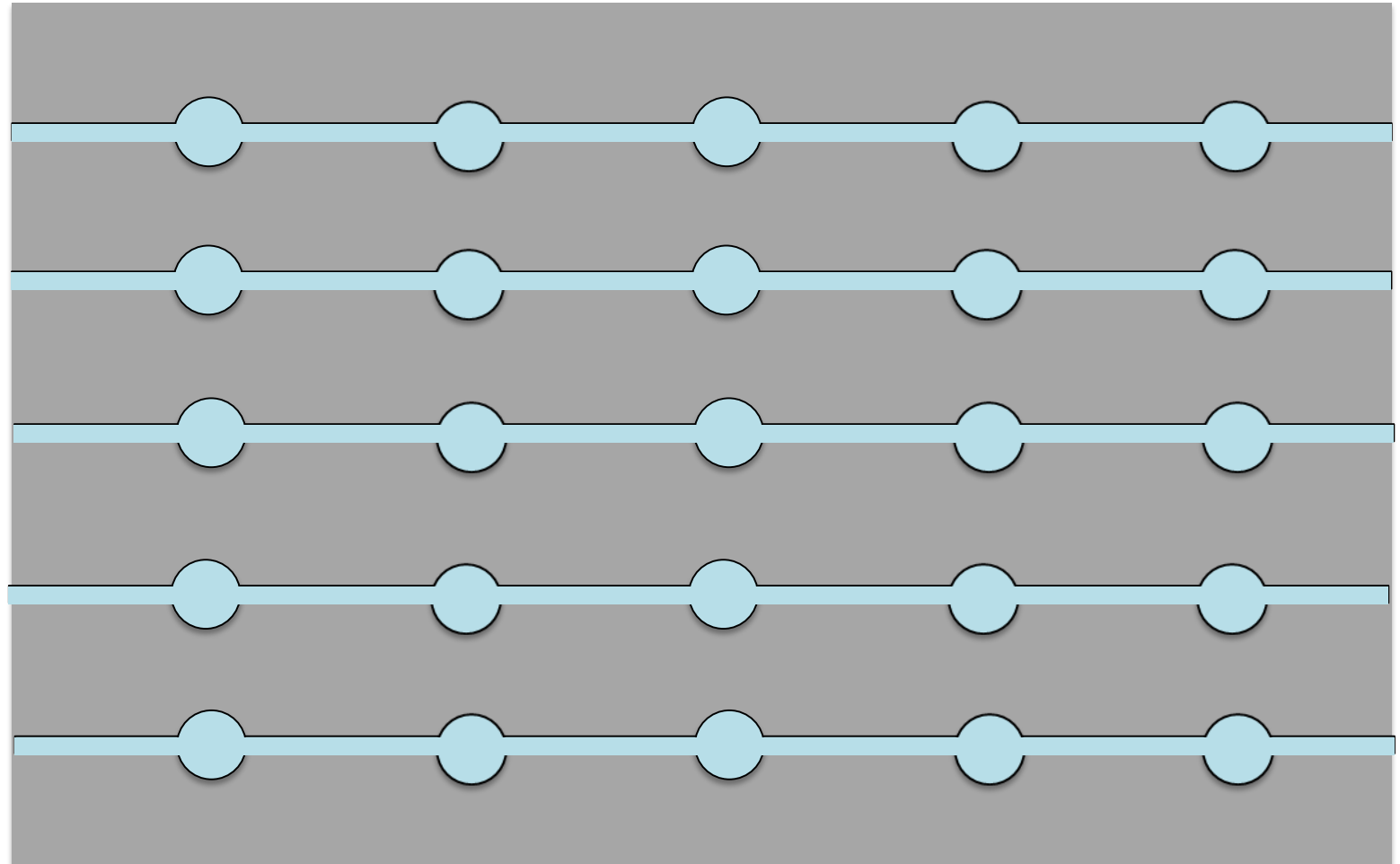


- $F = 5$

# Imagine a Chunk of Concrete



- 20% Porosity ( $\phi$ )
- 0.1 Connectivity ( $\beta$ )



- $F = 50$

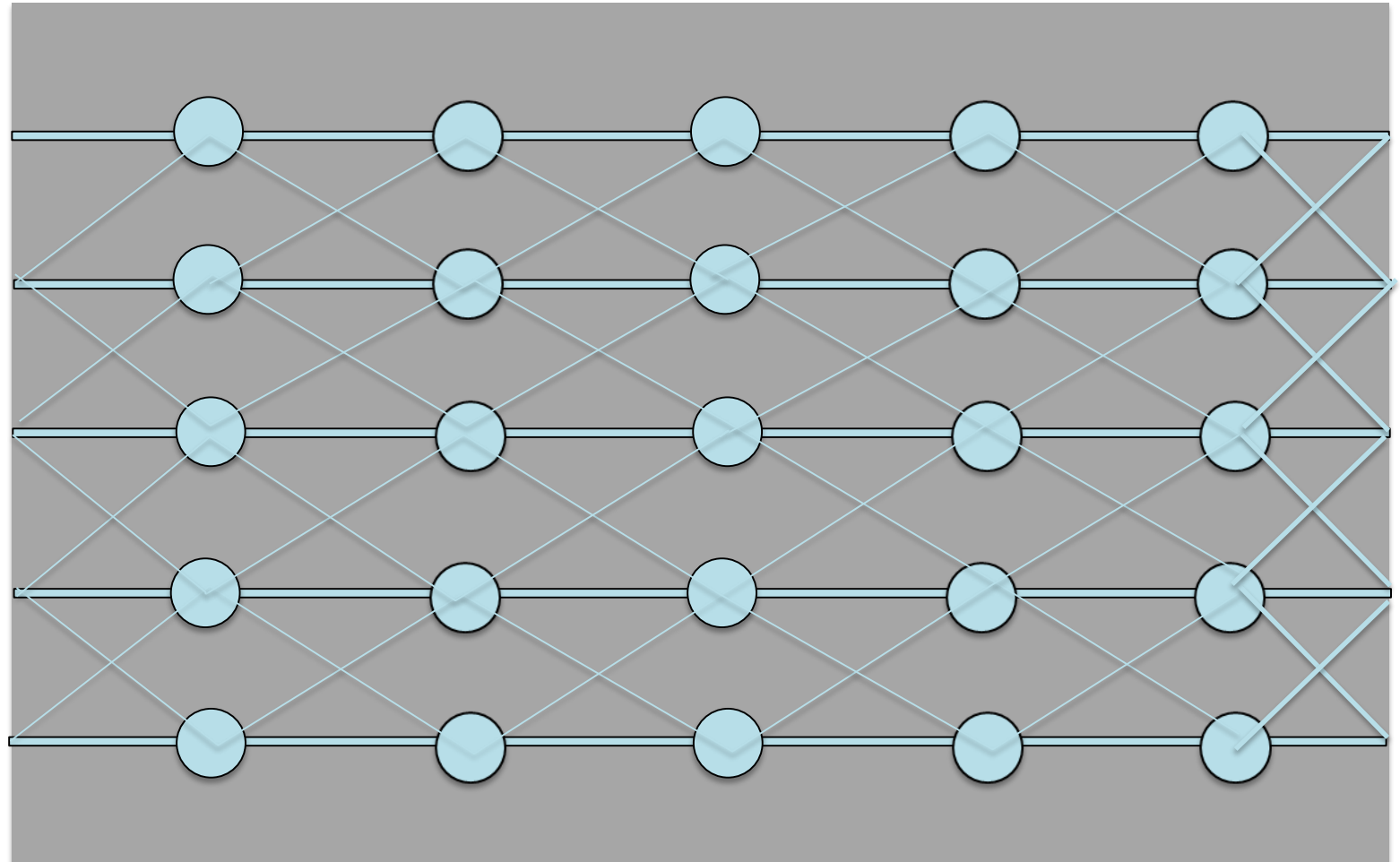


# Imagine a Chunk of Concrete



- 20% Porosity ( $\phi$ )
- 0.004 for a value of Connectivity ( $\beta$ )

Cl<sup>-</sup>

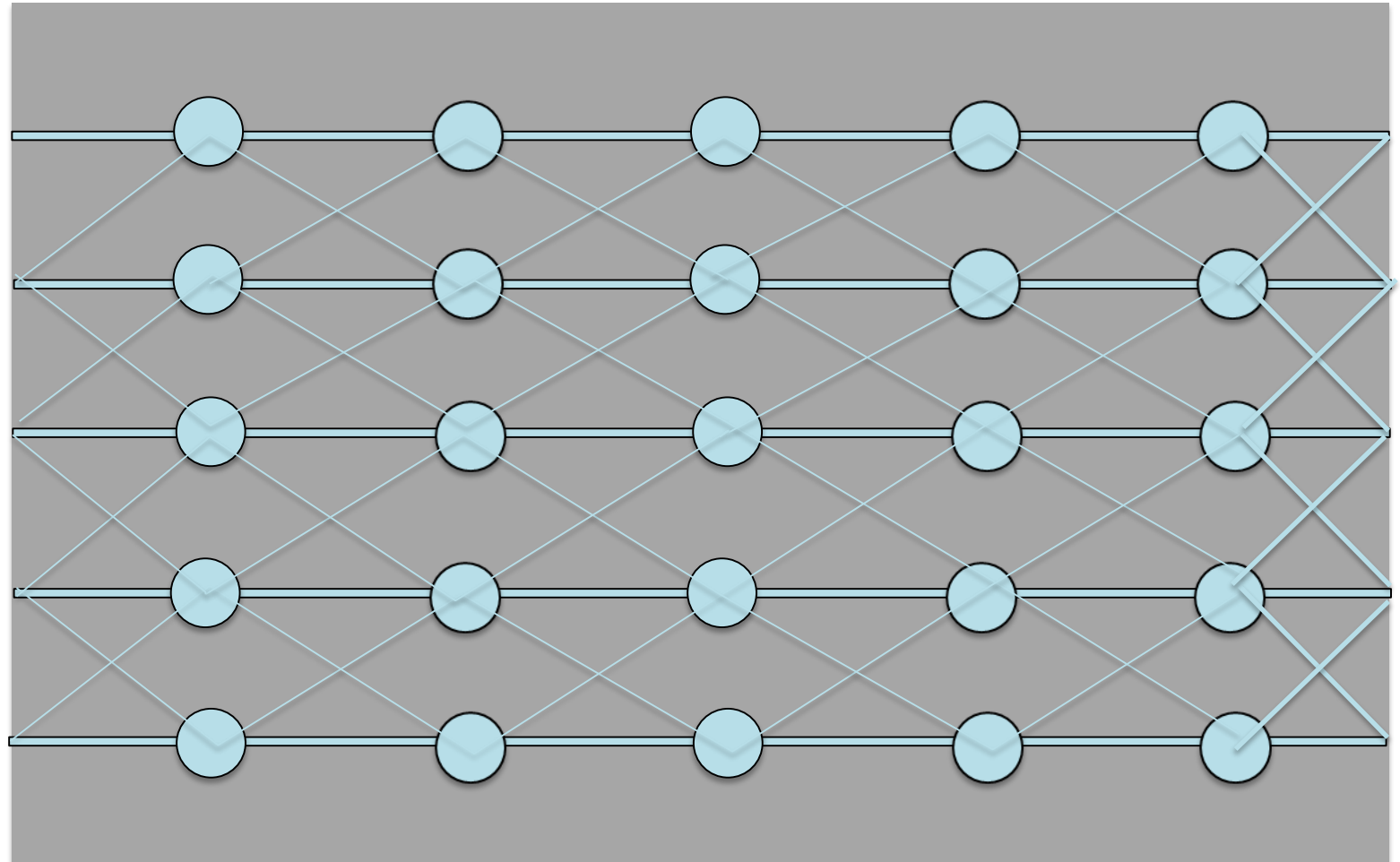


- $F = 1250$

# Change Alkali Content



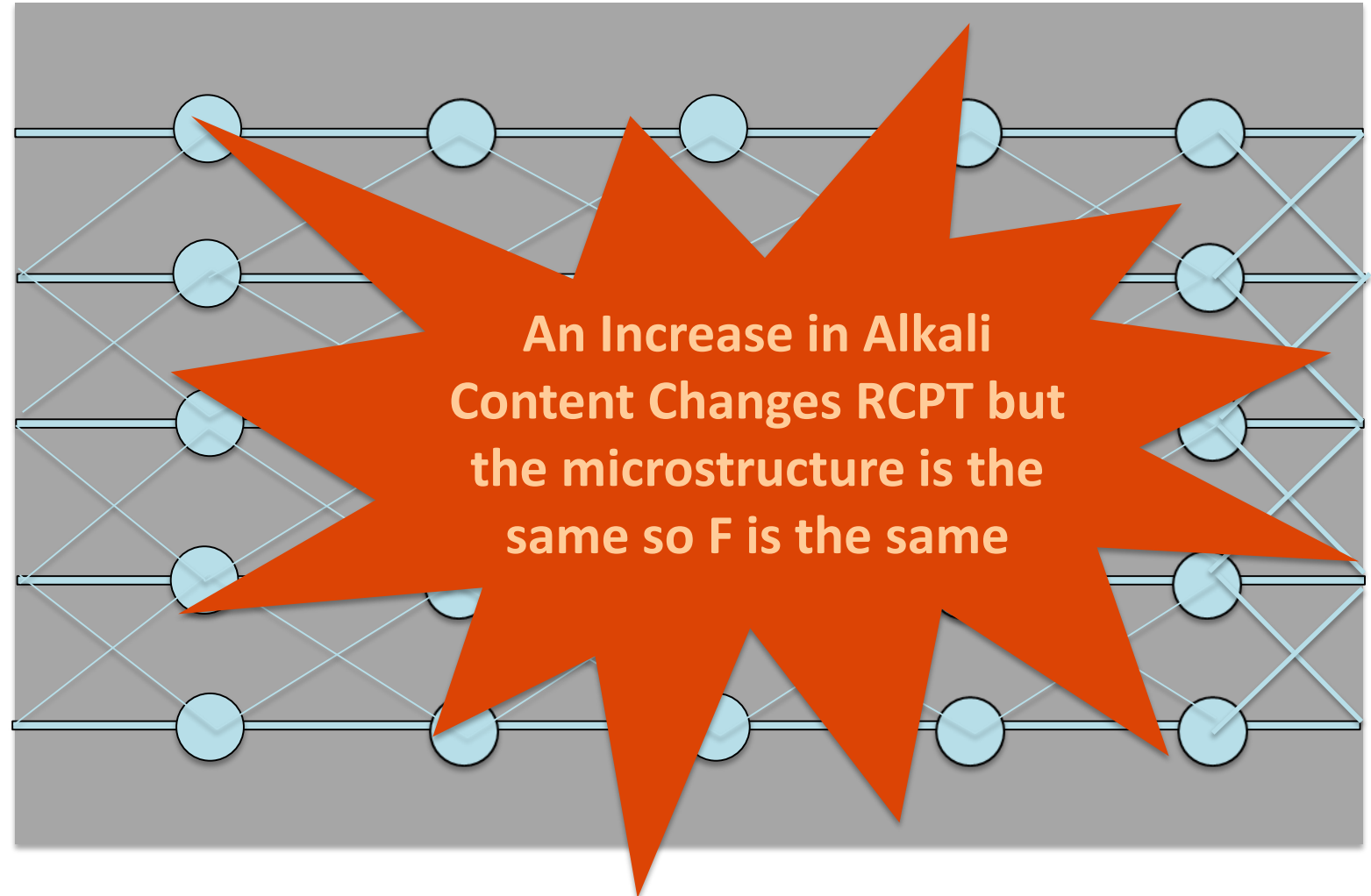
- 20% Porosity ( $\phi$ )
- 0.004 for a value of Connectivity ( $\beta$ )
- Double the alkali content of the cement
- $F = 1250$



# Change Alkali Content



- 20% Porosity ( $\phi$ )
- 0.004 for a value of Connectivity ( $\beta$ )
- Double the alkali content of the cement
- RCPT = 1250



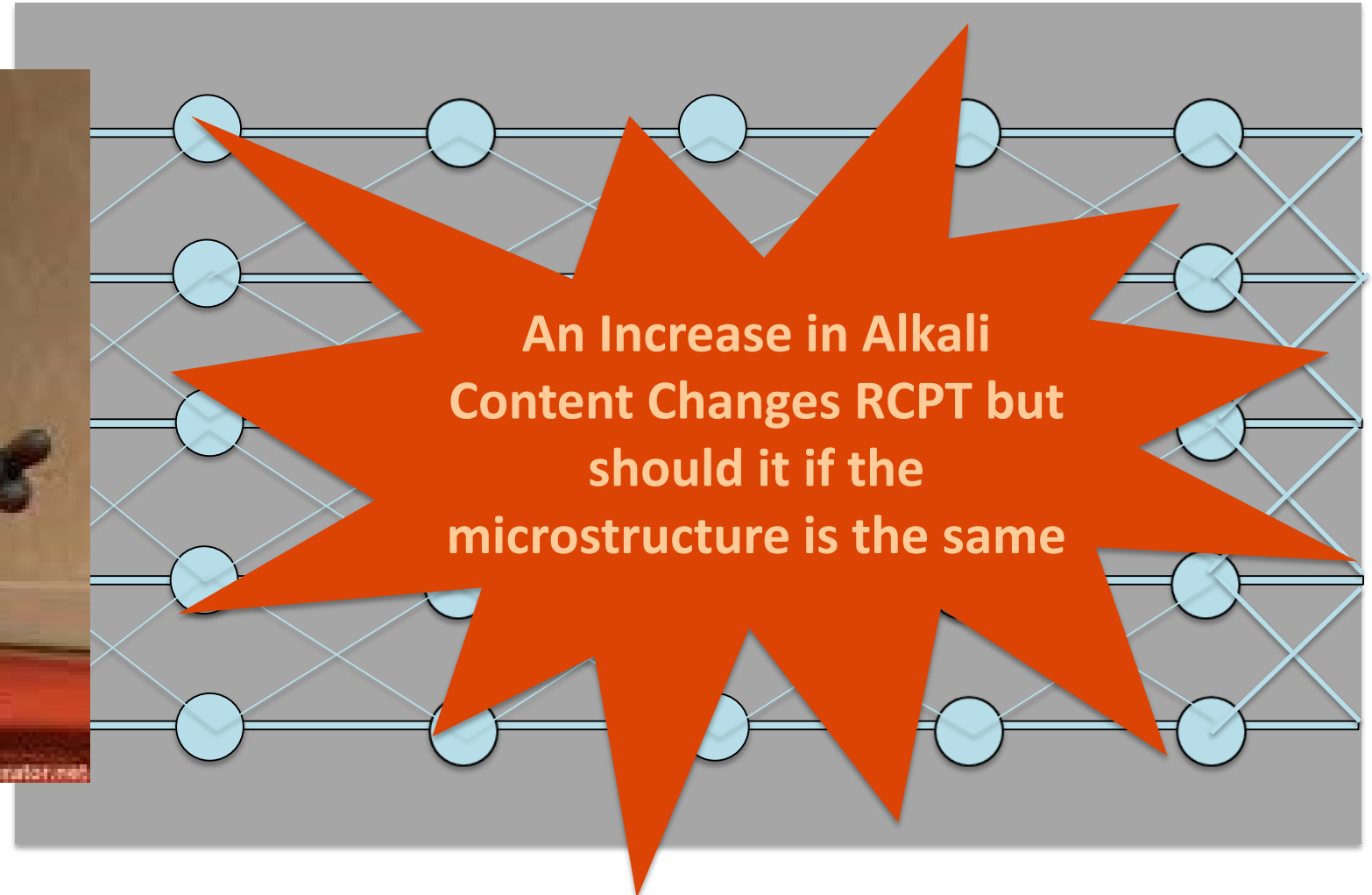
# Change Alkali Content



- 20% Porosity ( $\phi$ )



- RCPT = 3008



# Summary Slide



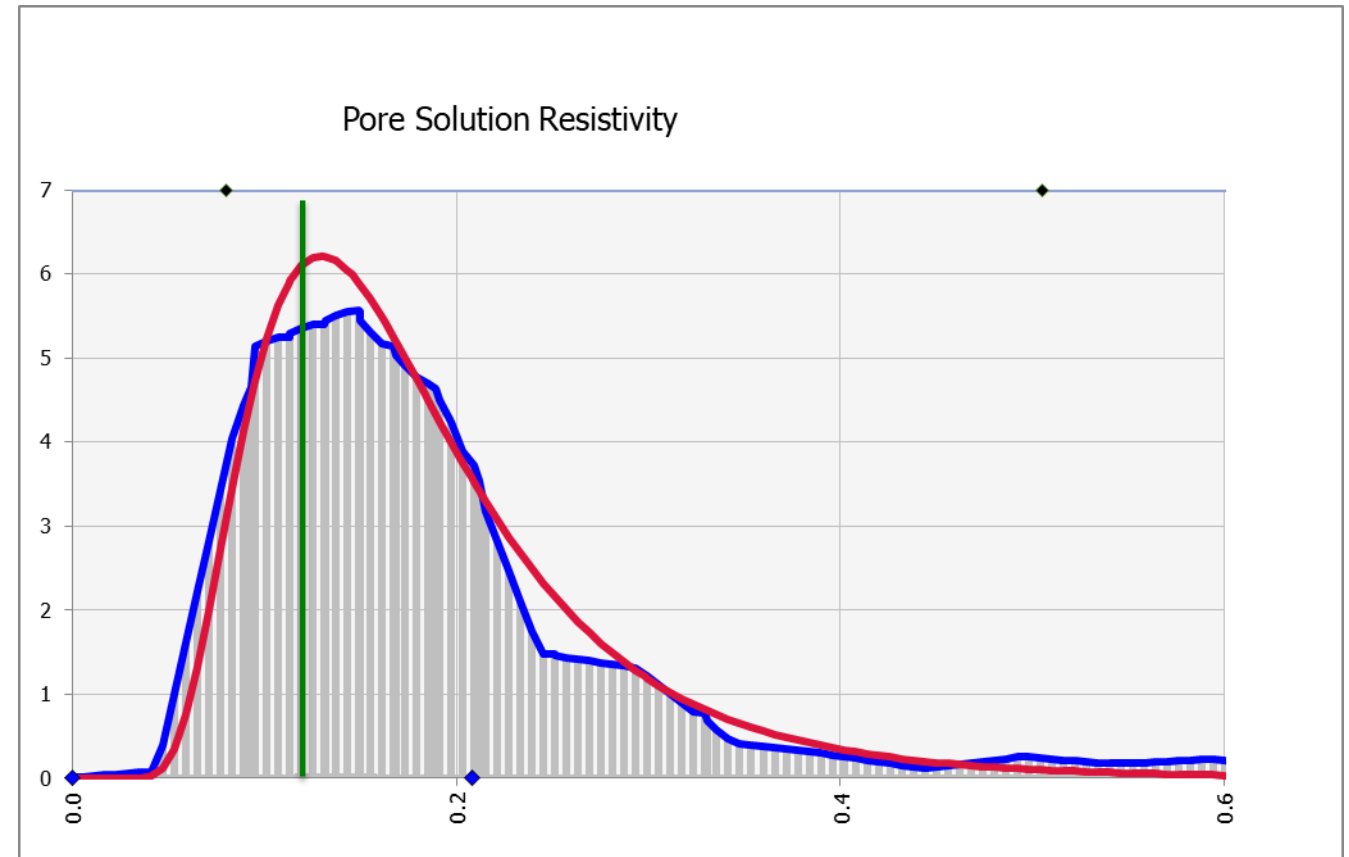
Case	RCPT	F Factor
Pure Non Conductive Solid	0	Infinite
20% Disconnected Porosity ( $\phi$ )	0	Infinite
20% Connected Porosity; Beta = 1	3760545	5
20% Con. Porosity; Beta = 0.1	3760	50
20% Con. Porosity; Beta = 0.004	1500	1250
20% Con. Porosity; Beta = 0.004 Double Alkali Content	3000	1250



# If we don't know anything about p. soln



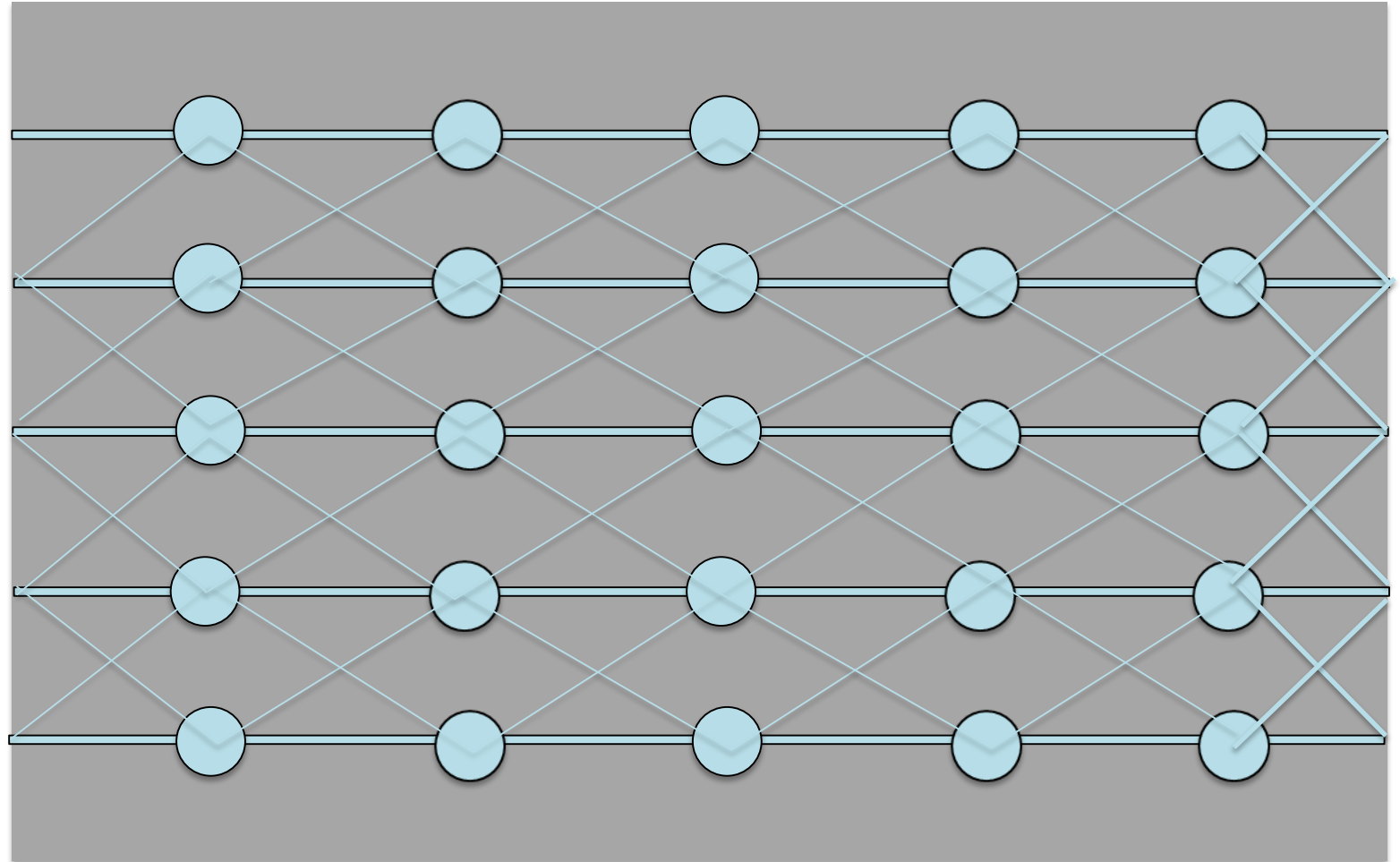
- A survey was done by L. Montanari about the pore solutions that have been reported in the literature (blue)
- Fitted data is shown in red (log normal)
- The typical resistivity we discuss as 'guessed' is 0.11 and is shown in green (its been an 'average value')



# Change Alkali Content



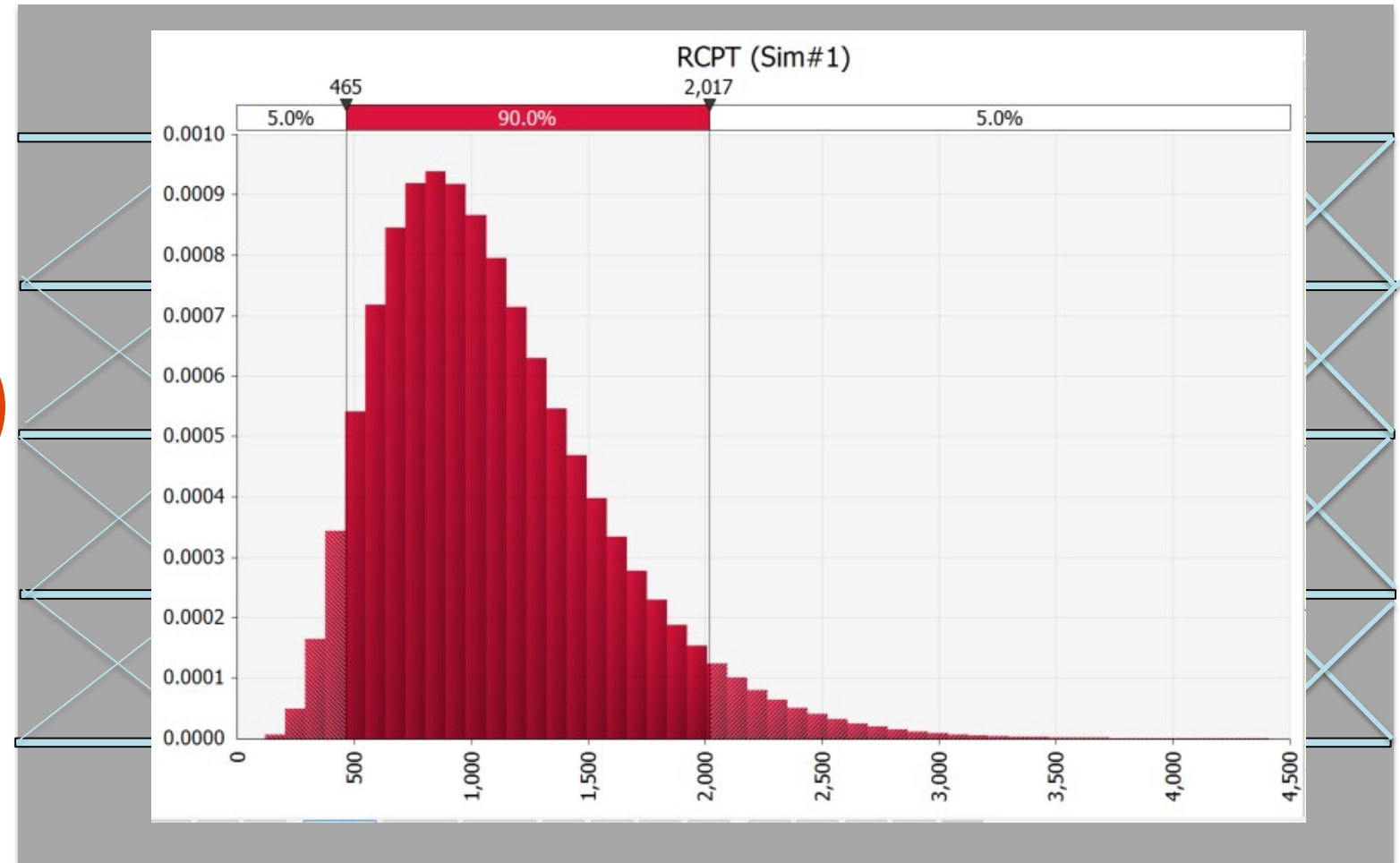
- 20% Porosity ( $\phi$ )
- 0.004 for a value of Connectivity ( $\beta$ )
- Lets use the pore solutions that we collected from the literature
- RCPT = ?



# Change Alkali Content



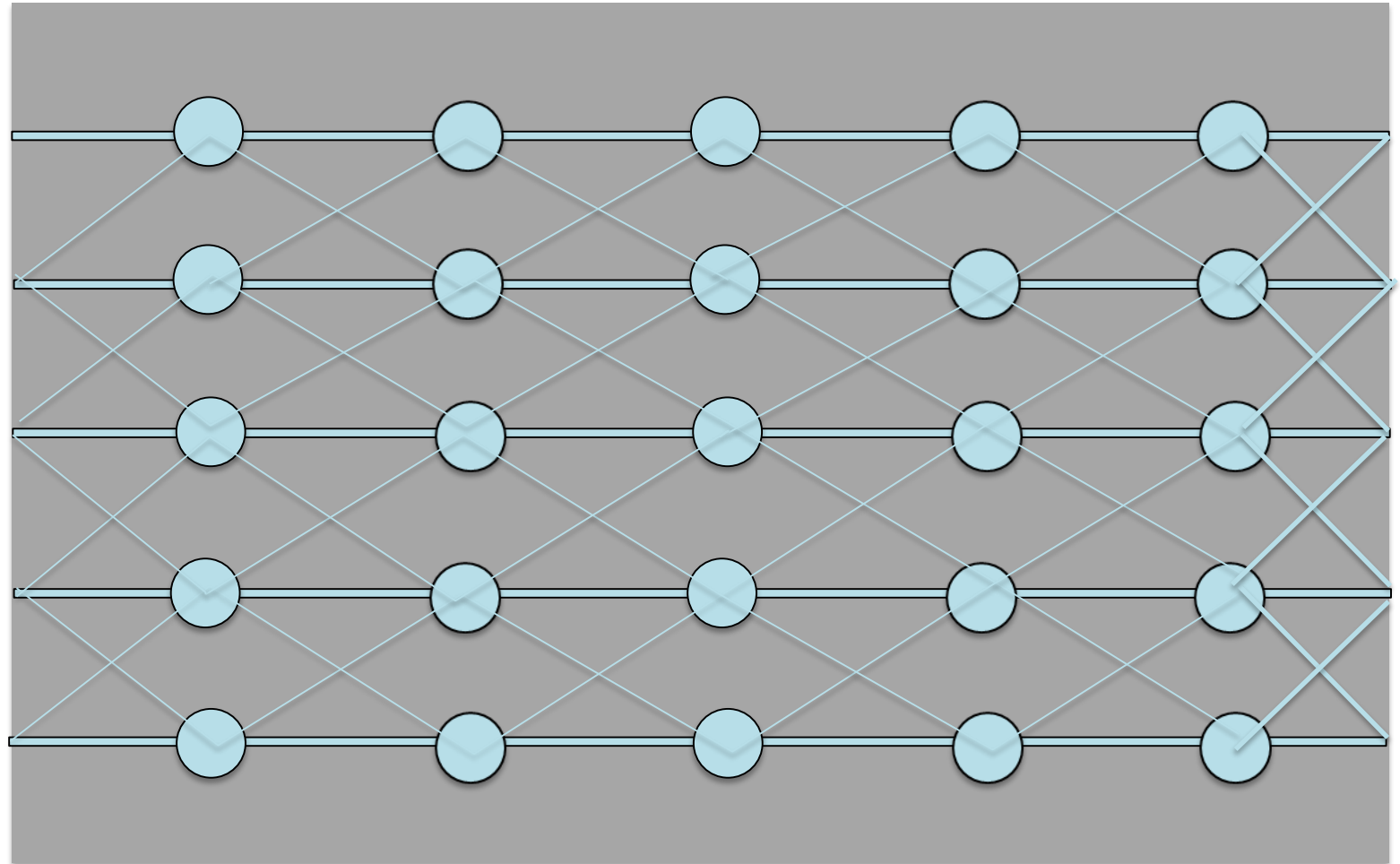
- 20% Porosity ( $\phi$ )
- 0.004 for a value of Connectivity ( $\beta$ )
- Lets use the pore solutions that we collected
- RCPT has a mean of 1100 w/ 90% CI limits 465 to 2017



# Change Alkali Content



- 20% Porosity ( $\phi$ )
- 0.004 for a value of Connectivity ( $\beta$ )
- Lets use the pore solutions that we collected from the literature
- $F = 1250$



# Summary Slide



Case	RCPT	F Factor
Pure Non Conductive Solid	0	Infinite
20% Disconnected Porosity ( $\phi$ )	0	Infinite
20% Connected Porosity; Beta = 1	3760545	5
20% Con. Porosity; Beta = 0.1	3760	50
20% Con. Porosity; Beta = 0.004	1500	1250
20% Con. Porosity; Beta = 0.004 Double Alkali Content	3000	1250
0% Con. Porosity; Beta = 0.004 Pore Solution Distribution	Varied	

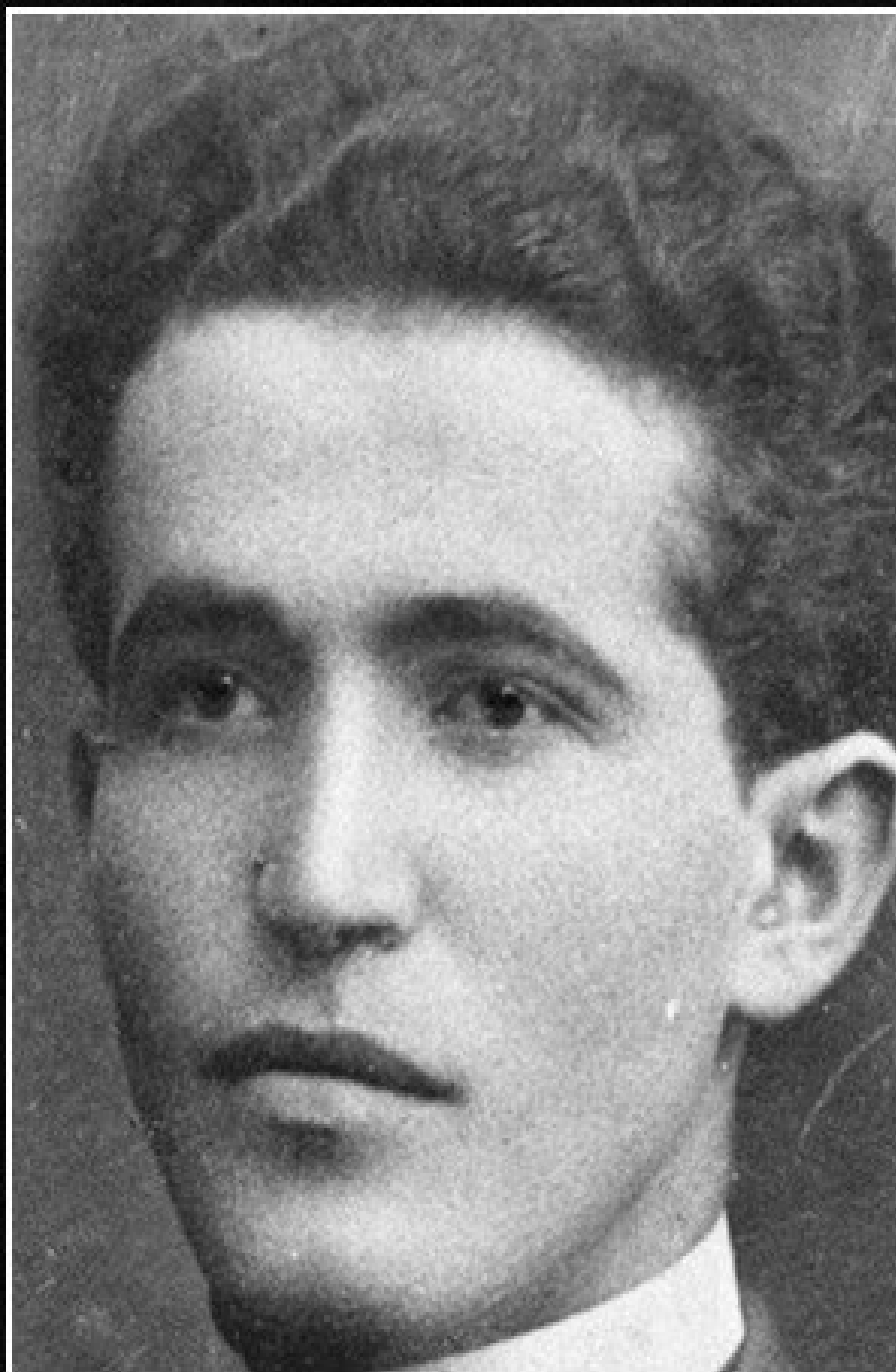


# Predicting Pore Solution



Oregon State University  
College of Engineering

- There are models that can be used to predict the ions in pore solution (these exist but depend on assumed solubility functions etc..)
- Once we know the composition we have a very strong algorithm to predict the resistivity
- Some have questioned the accuracy that we can expect to have with knowing the pore solution and speculated that this just cant be done



If an expert says it can't be done,  
get another expert.

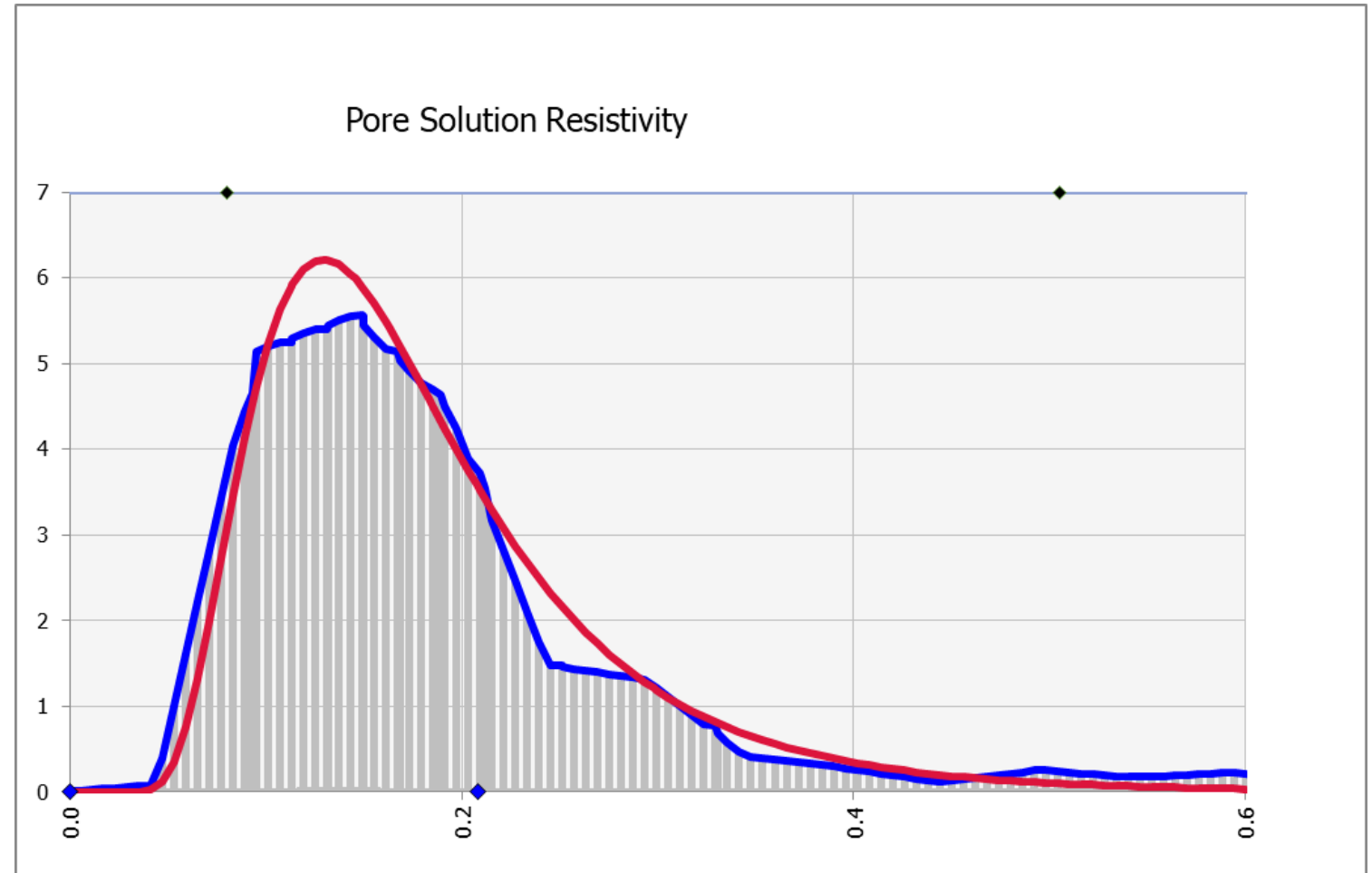
— *David Ben-Gurion* —

**AZ** QUOTES

# How Accurate Do We Need to Be with P Soln



- If we know nothing about the pore solution and just make a guess (0.11 ohm m)



# Monte Carlo Simulation



Oregon State University  
College of Engineering

- We can utilize the distributions for the pore solution and determine relative error
- Each case is simulated 10,000,000



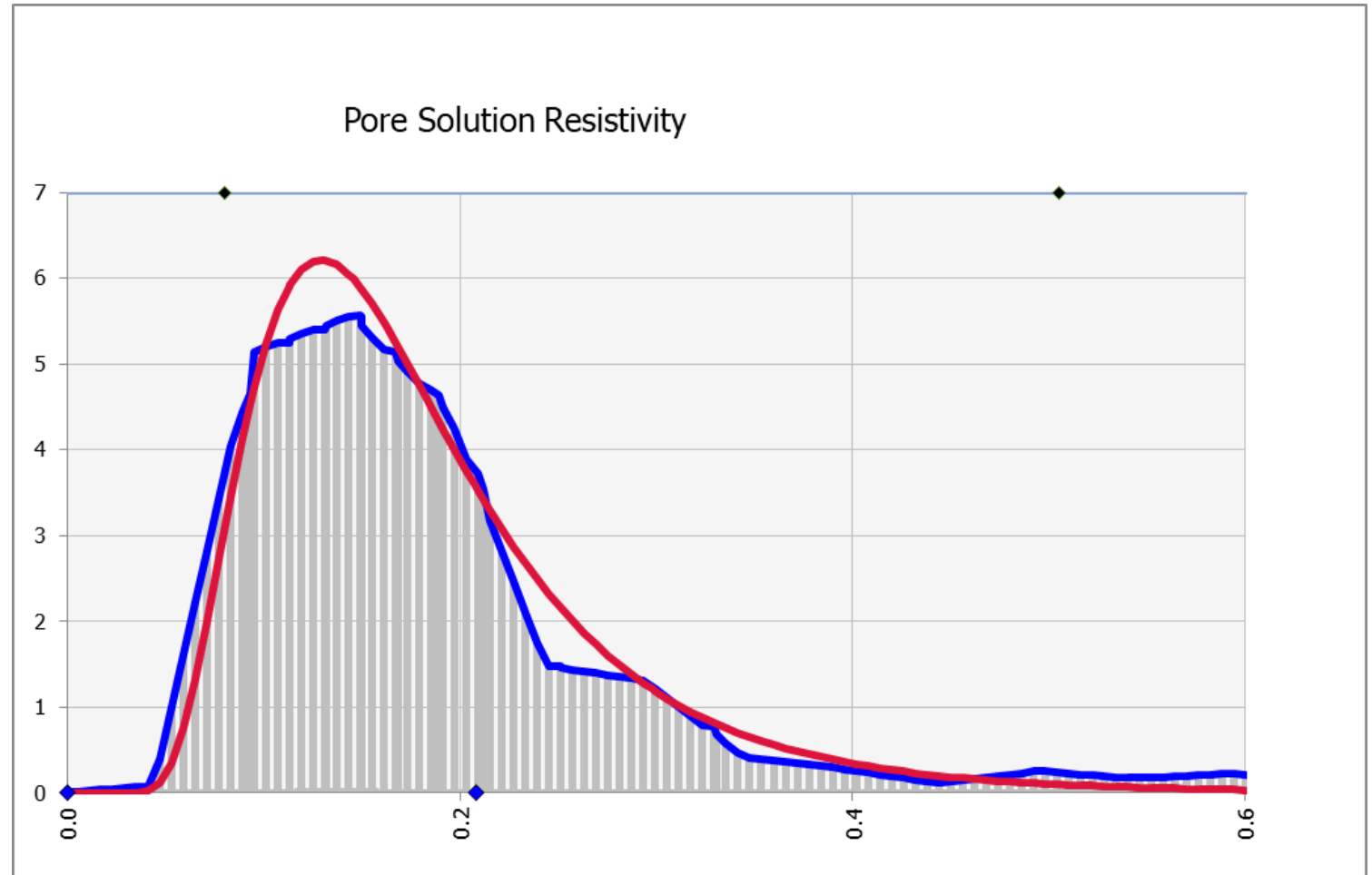
# How Accurate Do We Need to Be with P Soln



- If we know nothing about the pore solution and just make a guess (0.11 ohm m)

$$Estimate = \frac{F_{est} - F_{act}}{F_{est}}$$

- Estimate 0.84 (std dev)

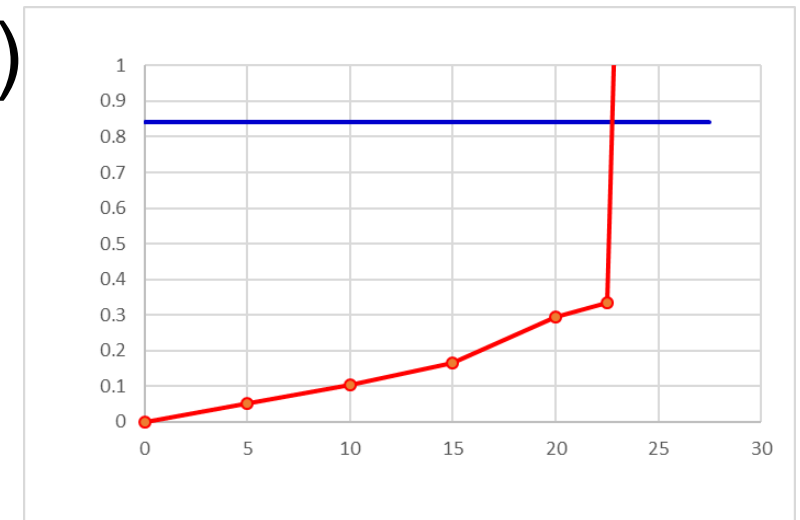




# How Accurate Do We Need to Be with P Soln



- Standard deviation of prediction knowing nothing about pore solution (shown in blue)
- We can estimate the pore solution using some algorithm with a known COV (error, shown on the x axis)
- It is only when the known COV (error) exceeds 22.5% that we are better off to just use a constant value
- As an example (+/- 2 SD)
- 0.11 would be 0.06 to 0.16



# RCPT is a solid tool



Oregon State University  
College of Engineering



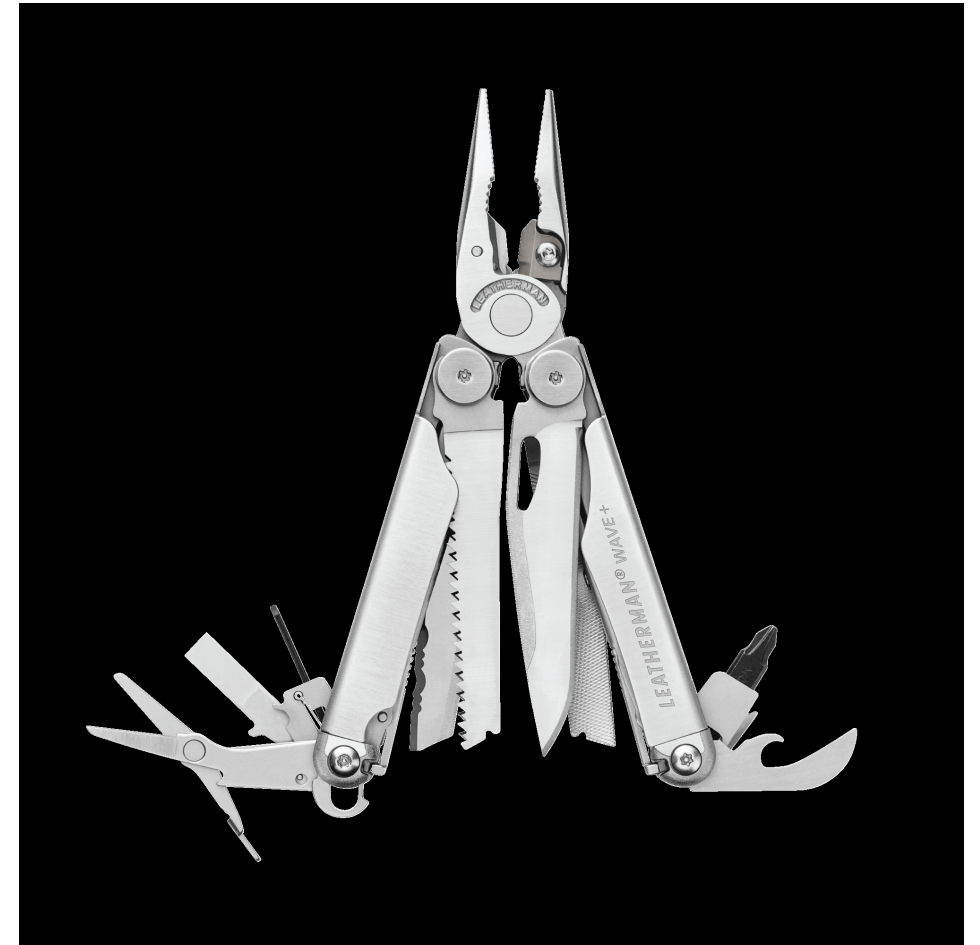
# Is this Versatile?



# Is this Versatile?



Oregon State University  
College of Engineering



# Formation Factor is the Leatherman of Transport



Oregon State University  
College of Engineering

- RCPT tells you RCPT and people feel comfortable
- The formation factor provides information that is directly related to several transport properties (Derivations available)
- Quality Control (water content; water to cement ratio)
- Ionic Diffusion (Effective Diffusion)
- Diffusion with Binding (Apparent Diffusion)
- Fluid Absorption (Sorptivity)
- Water/Hydraulic Permeability



# Shadow Projects



- It is important that as we do the shadow projects we perform the tests carefully
- We also need to carefully think about the work that needs to be done to advance the topic in a coordinated way as opposed to just repeating previous work

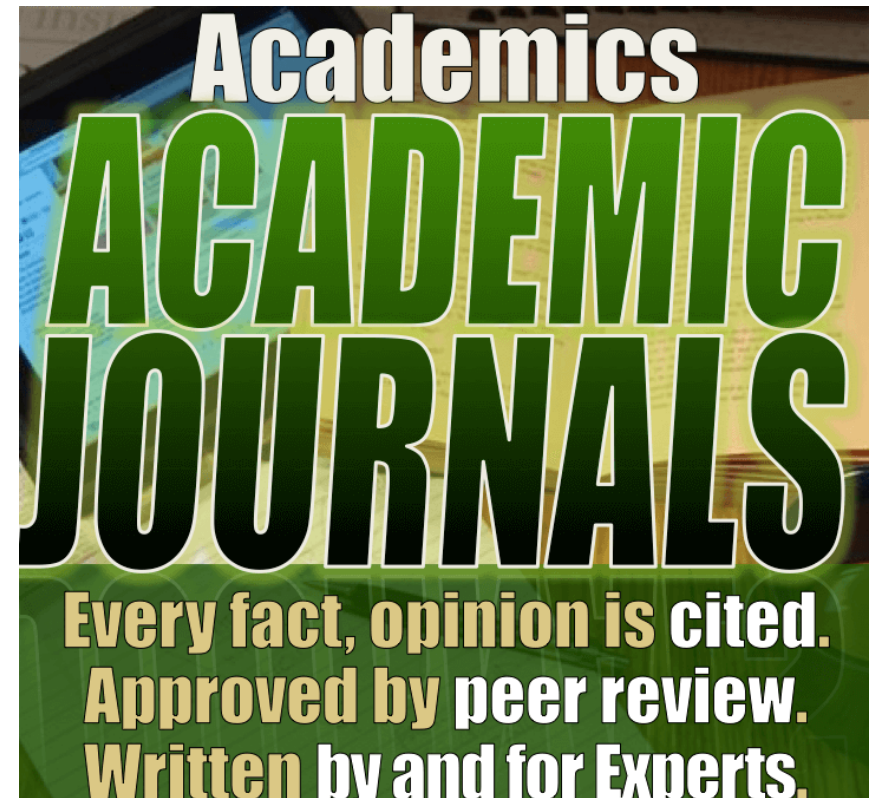


# Peer Reviewed Findings



Oregon State University  
College of Engineering

- In academic publishing, the goal of peer review is to assess the quality of articles submitted for publication in a scholarly journal.
- Many times there is speculation on what may or may not happen
- It would be good for us as a profession to document questions/concerns and then we can prove or disprove



# Summary



- We discussed how to perform the test (revised version of the test is with the AASHTO COMP committee)
- Option A – Bucket is an approach that is a compromise to have an easy approach and consistent pore solution
- We discussed thought experiments and showed how porosity, connectivity, and pore solution are related
- We have shown ‘accuracy needed’ for pore soln estimates
- We also know that formation factor describes a wide number of QC/QA and transport properties

# Use the Force



Oregon State University  
College of Engineering



# The F Factor and Bucket Test are in You SHA's



Oregon State University  
College of Engineering

